=> fil cap FILE 'CAPLUS' ENTERED AT 16:13:16 ON 15 JAN 2008 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2008 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 15 Jan 2008 VOL 148 ISS 3 FILE LAST UPDATED: 14 Jan 2008 (20080114/ED)

Effective October 17, 2005, revised CAS Information Use Policies apply. They are available for your review at:

http://www.cas.org/infopolicy.html

```
=> d que 119
                                       PLU=ON DEUTERATION+PFT/CT
          2077 SEA FILE=CAPLUS ABB=ON
L2
                                       PLU=ON DEUTERATION CATALYSTS+PFT,NT/CT
          287 SEA FILE=CAPLUS ABB=ON
L3
          2148 SEA FILE=CAPLUS ABB=ON PLU=ON L2 OR L3
L4
               TRANSFER PLU=ON L4 1- RN:
                                              22168 TERMS
L5
         22168 SEA FILE=REGISTRY ABB=ON PLU=ON L5
L6
           952 SEA FILE=REGISTRY ABB=ON PLU=ON L6 AND (PD OR PT OR RH OR RU
L7
               OR NI OR CO)/ELS
        154222 SEA FILE=CAPLUS ABB=ON PLU=ON L7(L)CAT+NT/RL
rs
           115 SEA FILE=CAPLUS ABB=ON PLU=ON L2 AND L3 AND L8
L10
               STR
L11
                           Ak~Cb Cb @8 O-^Ak @9 10
                   Ak @5
                  G4 18
```

VAR G1=5/6
VAR G2=5/8/6/9/11/OH
REP G3=(0-5) A
VAR G4=1/13
NODE ATTRIBUTES:
CONNECT IS E3 RC AT 1
CONNECT IS E1 RC AT 2
CONNECT IS E1 RC AT 5
CONNECT IS E2 RC AT 6

```
CONNECT IS E1 RC AT 7
CONNECT IS E1 RC AT 8
CONNECT IS E1 RC AT 10
CONNECT IS E1 RC AT 12
DEFAULT MLEVEL IS ATOM
GGCAT IS UNS AT 8
GGCAT IS UNS AT 12
DEFAULT ECLEVEL IS LIMITED
```

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE

L15 2492 SEA FILE=REGISTRY SUB=L6 SSS FUL L11

L16 1209 SEA FILE=REGISTRY ABB=ON PLU=ON L15 AND D/ELS L17 897 SEA FILE=CAPLUS ABB=ON PLU=ON L16(L)PREP+NT/RL

L19 22 SEA FILE=CAPLUS ABB=ON PLU=ON L17 AND L10

=> d 119 ibib abs hitind hitstr tot

L19 ANSWER 1 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2007:908108 CAPLUS Full-text

DOCUMENT NUMBER: 147:406255

TITLE: C-H bond activation by water on a palladium or

platinum metal surface

AUTHOR(S): Matsubara, Seijiro; Asano, Keisuke; Kajita, Yuichi;

Yamamoto, Mitsuru

CORPORATE SOURCE: Department of Material Chemistry, Graduate School of

Engineering, Kyoto University, Kyoudai-katsura, Kyoto,

606-8501, Japan

SOURCE: Synthesis (2007), (13), 2055-2059

CODEN: SYNTBF; ISSN: 0039-7881

PUBLISHER: Georg Thieme Verlag

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 147:406255

AB A water mol. is partially cleaved on a palladium or platinum metal surface under hydrothermal conditions to form an active platinum species. The species is effective for C-H bond functionalization which can be applied for H/D-exchange reactions, C-C bond-forming reactions, and C-N bond-forming reactions.

CC 22-4 (Physical Organic Chemistry)
 Section cross-reference(s): 66, 67

IT Deuteration

Deuteration catalysts

(C-H bond activation by water on a palladium or platinum metal surface)

IT 1314-08-5, Palladium oxide (PdO) 1314-15-4, Platinum oxide (PtO2) 3375-31-3 7440-05-3, Palladium, uses

7440-06-4, Platinum, uses 7440-44-0, Carbon, uses

7718-54-9, Nickel chloride, uses

RL: CAT (Catalyst use); USES (Uses)

(C-H bond activation by water on a palladium or platinum metal surface)

IT 111-67-1P, 2-Octene 112-40-3P, Dodecane 592-98-3P, 3-Octene

592-99-4P, 4-Octene 10249-89-5P 20617-93-0P, Quinoxaline-2,3,5,6,7,8-

d6 25378-22-7P, Dodecene 32190-42-4P 34071-94-8P, Quinoline-d7

36340-20-2P 73509-20-3P, 1H-Indole-1,2,3,4,5,6,7-d7 97797-70-1P

97960-58-2P 132125-39-4P 634897-78-2P 688320-42-5P 688320-43-6P

688320-44-7P 688320-45-8P 688320-46-9P **688320-48-1P**

```
880462-22-6P 951164-39-9P
                                   951164-40-2P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (C-H bond activation by water on a palladium or platinum metal surface)
IT
     7440-02-0, Raney nickel, uses
     RL: CAT (Catalyst use); USES (Uses)
        (catalysts; C-H bond activation by water on a palladium or platinum
       metal surface)
     1314-08-5, Palladium oxide (PdO) 1314-15-4, Platinum
ΙT
     oxide (PtO2) 3375-31-3 7440-05-3, Palladium, uses
     7440-06-4, Platinum, uses 7718-54-9, Nickel chloride,
     RL: CAT (Catalyst use); USES (Uses)
        (C-H bond activation by water on a palladium or platinum metal surface)
RN
     1314-08-5 CAPLUS
     Palladium oxide (PdO) (CA INDEX NAME)
CN
 0 = Pd
     1314-15-4 CAPLUS
RN
     Platinum oxide (PtO2) (CA INDEX NAME)
CN
 0===Pt===0
     3375-31-3 CAPLUS
RN
     Acetic acid, palladium(2+) salt (2:1) (CA INDEX NAME)
CN
  1/2 Pd(II)
     7440-05-3 CAPLUS
RN
     Palladium (CA INDEX NAME)
CN
 Pd
     7440-06-4 CAPLUS
RN
     Platinum (CA INDEX NAME)
CN
```

RN 7718-54-9 CAPLUS

CN Nickel chloride (NiCl2) (CA INDEX NAME)

Cl-Ni-Cl

IT 688320-48-1P

RL: SPN (Synthetic preparation); PREP (Preparation)

(C-H bond activation by water on a palladium or platinum metal surface)

RN 688320-48-1 CAPLUS

CN 2-Dodecanone-1,1,1,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12,12-d24 (CA INDEX NAME)

O || D3C-(CD2)9-C-CD3

IT **7440-02-0**, Raney nickel, uses

RL: CAT (Catalyst use); USES (Uses)

(catalysts; C-H bond activation by water on a palladium or platinum metal surface)

RN 7440-02-0 CAPLUS

CN Nickel (CA INDEX NAME)

Νi

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 2 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER:

2007:609303 CAPLUS Full-text

DOCUMENT NUMBER:

147:52633

TITLE:

Process for the homogeneous hydrogenation/deuteration of ketones into their corresponding secondary alkanols

using ruthenium catalytic systems

INVENTOR(S):

Heller, Detleff; Buschmann, Helmut; Drexler,

Hans-Joachim

PATENT ASSIGNEE(S):

Laboratorios Del Dr. Esteve, S.A., Spain

Eur. Pat. Appl., 11pp.

CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

SOURCE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.

KIND DATE

APPLICATION NO.

DATE

```
EP 2005-384038
                                                                    20051205
                                20070606
     EP 1792887
                          A1
         R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
             IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, AL,
             BA, HR, MK, YU
                                            WO 2006-EP69313
                                                                    20061205
     WO 2007065891
                          A1
                                20070614
            AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
             CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
             GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN,
             KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK,
             MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO,
             RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT,
             TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW
         RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
             IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ,
             CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH,
             GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
             KG, KZ, MD, RU, TJ, TM
                                            EP 2005-384038
                                                                A 20051205
PRIORITY APPLN. INFO.:
                         CASREACT 147:52633; MARPAT 147:52633
OTHER SOURCE(S):
     A process for the catalytic hydrogenation/deuteration of C1-6 alkyl ketones,
ΑB
     e.g., acetone, (including prochiral ketones) or their deuterated derivs. in
     high yield and selectivity into their resp. secondary alkanols, e.g.,
     isopropanol, is described. This process comprises using ruthenium achiral or
     chiral catalytic systems to yield racemic or nonracemic chiral alcs. or
     deuterated racemic or nonracemic chiral alcs. under high pressure and room
     temperature
     23-8 (Aliphatic Compounds)
CC
     Section cross-reference(s): 67
ΙT
     Deuteration
       Deuteration catalysts
     Hydrogenation
     Hydrogenation catalysts
        (process for the homogeneous hydrogenation/deuteration of ketones into
        their corresponding secondary alkanols using ruthenium catalytic
        systems)
     29841-69-8 134524-84-8 925941-06-6 939824-57-4
IT
     939824-58-5
     RL: CAT (Catalyst use); USES (Uses)
        (process for the homogeneous hydrogenation/deuteration of ketones into
        their corresponding secondary alkanols using ruthenium catalytic
        systems)
IT
     3976-29-2P, 2-Propan-1,1,1,3,3,3-d6-ol
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (process for the homogeneous hydrogenation/deuteration of ketones into
        their corresponding secondary alkanols using ruthenium catalytic
        systems)
     134524-84-8 925941-06-6 939824-57-4
IT
     939824-58-5
     RL: CAT (Catalyst use); USES (Uses)
        (process for the homogeneous hydrogenation/deuteration of ketones into
        their corresponding secondary alkanols using ruthenium catalytic
        systems)
     134524-84-8 CAPLUS
RN
     Ruthenium, [(1S)-[1,1'-binaphthalene]-2,2'-diylbis[diphenylphosphine-
CN
```

(CA INDEX NAME)

 κP]]dichloro-, (SP-4-2)- (9CI)

RN 925941-06-6 CAPLUS

CN Ruthenium, [1,1'-(1S)-[1,1'-binaphthalene]-2,2'-diylbis[1,1-diphenylphosphine-κP]][(1S,2S)-1,2-diphenyl-1,2-ethanediamine-κN1,κN2]hydro[tetrahydroborato(1-)-κH]-, (OC-6-22)- (CA INDEX NAME)

RN 939824-57-4 CAPLUS

CN Ruthenium, $[1,1'-[1,1'-binaphthalene]-2,2'-diylbis[1,1-diphenylphosphine-<math>\kappa P]$]hydro[tetrahydroborato(1-)- κH]- (CA INDEX NAME)

RN 939824-58-5 CAPLUS

CN Ruthenium, dichloro[2-(dicyclohexylphosphino-κP)-1-[1-(dicyclohexylphosphino-κP)-2-naphthalenyl]-5-methyl-1H-pyrrole]-

(CA INDEX NAME)

PAGE 1-A

PAGE 2-A



3976-29-2P, 2-Propan-1,1,1,3,3,3-d6-ol IT

RL: SPN (Synthetic preparation); PREP (Preparation)

(process for the homogeneous hydrogenation/deuteration of ketones into their corresponding secondary alkanols using ruthenium catalytic systems)

3976-29-2 CAPLUS RN

CN 2-Propan-1,1,1,3,3,3-d6-ol (9CI) (CA INDEX NAME)

D3C-CH-CD3

REFERENCE COUNT:

4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 3 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:1014792 CAPLUS Full-text

DOCUMENT NUMBER:

146:7638

TITLE:

C-H/C-D exchange reactions of aromatic compounds in

```
10/539,188
                         D2O with NaBD4-activated catalysts
AUTHOR(S):
                         Derdau, Volker; Atzrodt, Jens
CORPORATE SOURCE:
                         GMPK, Isotope Chemistry & Metabolite Synthesis
                         Frankfurt, Sanofi-Aventis Deutschland GmbH,
                         Frankfurt/Hoechst, 65926, Germany
SOURCE:
                         Synlett (2006), (12), 1918-1922
                         CODEN: SYNLES; ISSN: 0936-5214
PUBLISHER:
                         Georg Thieme Verlag
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
OTHER SOURCE(S):
                         CASREACT 146:7638
     A safe and efficient method for catalytic H/D exchange to provide high
     deuterium incorporation into a variety of aromatic substrates was developed.
     Systematic screening of the catalyst and activator revealed that the essential
     activation of the Pd catalyst could be achieved under safe and user friendly
     conditions. The application of this simple catalytic method for the
     deuteration of bi- and tricyclic aromatic compds. and chiral natural products
     was investigated.
     25-2 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)
     Section cross-reference(s): 27
    Deuteration
TΤ
      Deuteration catalysts
        (C-H/C-D exchange reactions of aromatic compds. in D2O with
       NaBD4-activated catalysts)
     7440-05-3, Palladium, uses 7647-10-1, Palladium(II)
IT
     chloride
     RL: CAT (Catalyst use); USES (Uses)
        (C-H/C-D exchange reactions of aromatic compds. in D2O with
       NaBD4-activated catalysts)
     37464-79-2P
                  651316-73-3P
IT
                                  915232-04-1P
                                                 915232-06-3P
     915232-14-3P
                   915232-16-5P
                                   915232-18-7P
                                                  915232-20-1P
                                                                  915232-22-3P
     915232-24-5P
                    915232-26-7P
                                   915232-28-9P
                                                  915232-30-3P
                                                                  915232-33-6P
     915232-35-8P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (C-H/C-D exchange reactions of aromatic compds. in D2O with
       NaBD4-activated catalysts)
IT
     7440-05-3, Palladium, uses 7647-10-1, Palladium(II)
     chloride
     RL: CAT (Catalyst use); USES (Uses)
        (C-H/C-D exchange reactions of aromatic compds. in D2O with
       NaBD4-activated catalysts)
    7440-05-3 CAPLUS
RN
CN
    Palladium (CA INDEX NAME)
Pd
```

```
7647-10-1 CAPLUS
RN
CN
    Palladium chloride (PdCl2) (CA INDEX NAME)
```

Cl-Pd-Cl

RL: SPN (Synthetic preparation); PREP (Preparation)

(C-H/C-D exchange reactions of aromatic compds. in D20 with NaBD4-activated catalysts)

RN 37464-79-2 CAPLUS

CN Benzenemethan-d-ol, α-propyl- (9CI) (CA INDEX NAME)

REFERENCE COUNT:

58 THERE ARE 58 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 4 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER:

2006:689620 CAPLUS Full-text

DOCUMENT NUMBER:

146:421688

TITLE:

Synergistic effect of a palladium-on-carbon/platinum-

on-carbon mixed catalyst in hydrogen/deuterium exchange reactions of alkyl-substituted aromatic

compounds

AUTHOR(S):

Ito, Nobuhiro; Watahiki, Tsutomu; Maesawa, Tsuneaki;

Maegawa, Tomohiro; Sajiki, Hironao

CORPORATE SOURCE:

Chemical Products Research Laboratories, Wako Pure

Chemical Industries, Ltd., 1633 Matoba, Kawagoe,

350-1101, Japan

SOURCE:

Advanced Synthesis & Catalysis (2006), 348(9),

1025-1028

CODEN: ASCAF7; ISSN: 1615-4150

PUBLISHER:

Wiley-VCH Verlag GmbH & Co. KGaA

DOCUMENT TYPE:

Journal LANGUAGE: English

A synergistic effect in the H-D exchange reaction of alkyl-substituted aromatic compds. using the Pd/C-Pt/C-D2O-H2 system was discovered. This system would lead to fully H-D exchange results even on the sterically hindered sites which were only low-deuterium incorporated by Pd/C or Pt/C independently. Since the reaction was general for a variety of aromatic compds., it could be applied to the deuteration of dianiline derivs. as raw materials for polyimides.

25-17 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds) CC

ΙT Deuteration

Deuteration catalysts

Exchange reaction

Exchange reaction catalysts

(synergistic effect of palladium-on-carbon/platinum-on-carbon mixed catalyst in hydrogen/deuterium exchange reaction of alkyl-substituted aromatic compds.)

7440-05-3, Palladium, uses 7440-06-4, Platinum, uses IT

RL: CAT (Catalyst use); USES (Uses)

(synergistic effect of palladium-on-carbon/platinum-on-carbon mixed catalyst in hydrogen/deuterium exchange reaction of alkyl-substituted aromatic compds.)

767627-97-4P, Benzene-d5-pentanoic-d8 acid IT 861405-62-1P

870284-60-9P 870284-54-1P 870284-63-2P 870284-66-5P 870284-69-8P

934266-51-0P 934266-52-1P **934266-54-3P**

RL: SPN (Synthetic preparation); PREP (Preparation)

(synergistic effect of palladium-on-carbon/platinum-on-carbon mixed catalyst in hydrogen/deuterium exchange reaction of alkyl-substituted aromatic compds.)

IT 7440-05-3, Palladium, uses 7440-06-4, Platinum, uses

RL: CAT (Catalyst use); USES (Uses)

(synergistic effect of palladium-on-carbon/platinum-on-carbon mixed catalyst in hydrogen/deuterium exchange reaction of alkyl-substituted aromatic compds.)

RN 7440-05-3 CAPLUS

CN Palladium (CA INDEX NAME)

Pd

RN 7440-06-4 CAPLUS

CN Platinum (CA INDEX NAME)

Рt

IT 767627-97-4P, Benzene-d5-pentanoic-d8 acid 934266-54-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(synergistic effect of palladium-on-carbon/platinum-on-carbon mixed catalyst in hydrogen/deuterium exchange reaction of alkyl-substituted aromatic compds.)

RN 767627-97-4 CAPLUS

CN Benzene-d5-pentanoic-d8 acid (CA INDEX NAME)

$$D \longrightarrow D \qquad (CD_2)_{4-CO_2H}$$

RN 934266-54-3 CAPLUS

CN Benzene-2,3,4,5-d4-pentanoic- α , α , β , β , γ , γ , δ , δ -d8 acid (CA INDEX NAME)

REFERENCE COUNT:

42 THERE ARE 42 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

```
L19 ANSWER 5 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN
ACCESSION NUMBER:
                         2006:108991 CAPLUS Full-text
DOCUMENT NUMBER:
                          144:292970
TITLE:
                          Synthesis of base-selectively deuterium-labeled
                          nucleosides by the pd/C-catalyzed H-D exchange
                          reaction in deuterium oxide
AUTHOR(S):
                          Esaki, Hiroyoshi; Aoki, Fumiyo; Maegawa, Tomohiro;
                         Hirota, Kosaku; Sajiki, Hironao
                         Laboratory of Medicinal Chemistry, Gifu Pharmaceutical
CORPORATE SOURCE:
                         University, Mitahora-higashi, Gifu, 502-8585, Japan
SOURCE:
                         Heterocycles (2005), 66, 361-369
                         CODEN: HTCYAM; ISSN: 0385-5414
PUBLISHER:
                         Japan Institute of Heterocyclic Chemistry
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
OTHER SOURCE(S):
                         CASREACT 144:292970
AΒ
     The D2 gas-free and base-selective H-D exchange reaction of nucleosides was
     developed. It discloses a convenient route to the post-synthetic
     incorporation of deuteriums into the base moiety of nucleic acids with high
     deuterium efficiency.
CC
     33-9 (Carbohydrates)
ΙT
     Deuteration
        (regioselective; synthesis of base-selectively deuterium-labeled
        nucleosides by the pd/C-catalyzed H-D exchange reaction in deuterium
        oxide)
   Deuteration catalysts
        (synthesis of base-selectively deuterium-labeled nucleosides by the
        pd/C-catalyzed H-D exchange reaction in deuterium oxide)
IT
     7440-05-3, Palladium, uses
     RL: CAT (Catalyst use); USES (Uses)
        (synthesis of base-selectively deuterium-labeled nucleosides by the
        pd/C-catalyzed H-D exchange reaction in deuterium oxide)
ΙT
     24897-52-7P, 2,4(1H,3H)-Pyrimidinedione-5,6-d2 28671-50-3P,
     Cytidine-5-d 40436-51-9P, Uridine-5-d 40632-21-1P,
     Uridine-5,6-d2 40632-25-5P, Cytidine-5,6-d2 74848-84-3P
     , Thymidine-\alpha, \alpha, \alpha-d3 82845-88-3P.
     Adenosine-2,8-d2
                       90742-80-6P 96412-41-8P, Guanosine-8-d
     102147-86-4P
                    106391-24-6P
                                   200496-79-3P
                                                   697807-00-4P,
     1H-Purin-2,8-d2-6-amine 697807-01-5P, Inosine-2,8-d2
     697807-02-6P 860788-48-3P 879005-77-3P,
     Thymidine-\alpha-d 879005-78-4P, Thymidine-\alpha, \alpha-d2
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (synthesis of base-selectively deuterium-labeled nucleosides by the
        pd/C-catalyzed H-D exchange reaction in deuterium oxide)
IT
     7440-05-3, Palladium, uses
     RL: CAT (Catalyst use); USES (Uses)
        (synthesis of base-selectively deuterium-labeled nucleosides by the
```

pd/C-catalyzed H-D exchange reaction in deuterium oxide)

Pd

RN CN 7440-05-3 CAPLUS

Palladium (CA INDEX NAME)

28671-50-3P, Cytidine-5-d 40436-51-9P, Uridine-5-d IT40632-21-1P, Uridine-5,6-d2 40632-25-5P, Cytidine-5,6-d2 **74848-84-3P**, Thymidine- α , α , α -d3 82845-88-3P, Adenosine-2,8-d2 96412-41-8P, Guanosine-8-d 697807-01-5P, Inosine-2,8-d2 860788-48-3P 879005-77-3P, Thymidine- α -d 879005-78-4P, Thymidine- α , α -d2 RL: SPN (Synthetic preparation); PREP (Preparation) (synthesis of base-selectively deuterium-labeled nucleosides by the pd/C-catalyzed H-D exchange reaction in deuterium oxide) 28671-50-3 CAPLUS RN Cytidine-5-d (8CI, 9CI) CN (CA INDEX NAME)

Absolute stereochemistry.

RN 40436-51-9 CAPLUS CN Uridine-5-d (9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 40632-21-1 CAPLUS CN Uridine-5,6-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

RN 40632-25-5 CAPLUS CN Cytidine-5,6-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

RN 74848-84-3 CAPLUS

CN Thymidine- α , α , α -d3 (9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 82845-88-3 CAPLUS

CN Adenosine-2,8-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

RN 96412-41-8 CAPLUS

CN Guanosine-8-d (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

RN 697807-01-5 CAPLUS

CN Inosine-2,8-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

RN 860788-48-3 CAPLUS

CN Adenosine-2,8-d2, 2'-deoxy- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

RN 879005-77-3 CAPLUS

CN Thymidine- α -d (9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 879005-78-4 CAPLUS

CN Thymidine- α , α -d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT: 32 THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 6 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2005:696848 CAPLUS Full-text

DOCUMENT NUMBER: 143:172769

TITLE: Method of deuteration of aromatic ring and/or

heterocycle compounds using mixed metal catalyst

INVENTOR(S): Ito, Nobuhiro; Maesawa, Tsuneaki; Muto, Kazushige;

Hirota, Kosaku; Sajiki, Hironao

PATENT ASSIGNEE(S): Wako Pure Chemical Industries, Ltd., Japan

SOURCE: PCT Int. Appl., 55 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

P

	PAT	CENT	NO.			KIN	D 	DATE		-						D	ATE	
	WO 2005070853			A1		2005	0804	WO 2004-JP19049										
		W:	ΑE,	AG,	AL,	AM,	ΑT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	ΒZ,	CA,	CH,
																		GD,
																		LC,
								LV,										
								PL,										
								TZ,										
		RW:	BW,															
			ΑZ,	BY,	KG,	ΚŻ,	MD,	RU,	ΤJ,	TM,	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,
								GR,										
								BF,										
						TD,												
	CA	2553	376			A 1		2005	0804	(CA 20	004-2	2553	376		21	0041	221
	ΕP	1707	548			A 1		2006	1004]	EP 20	004-8	8074	06		2	0041	221
		R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,
			ΙE,	SI,	LT,	FI,	RO,	CY,	TR,	BG,	CZ,	EE,	HU,	PL,	SK,	IS		
	CN	1906	143			Α		2007	0131	(CN 20	004-8	3004	0874		20	0041	221
RIOR	IORITY APPLN. INFO.: JP 2004-16075 A 20040123																	
										1	WO 20	004-	JP190	049	V	v 20	0041	221
D .	70		ء د		. 		2	- 1	1				1					

AB A method of deuteration in which a compound with aromatic ring and/or heterocycle having an enhanced deuteration ratio can be obtained. There is provided a method of deuterating a compound with aromatic ring and/or

heterocycle, characterized in that a compound with aromatic ring and/or heterocycle is reacted with a deuterium source in the presence of an activated mixed catalyst composed of at least two members selected from among a palladium catalyst, a platinum catalyst, a rhodium catalyst, an iridium catalyst, a ruthenium catalyst, a nickel catalyst and a cobalt catalyst. Thus, 500 mg nicotinic acid, 50 mg Pd/C (5 mg Pd), and 100 mg Pt/C (5 mg Pt) were suspended in 17 mL D2O, sealed, purged with H, and heated at 180° for .apprx.24 h to give deuterated nicotinic acid with 99% deuteration at 2, 5, and 6 positions and 48% deuteration at 4 position vs. 98% deuteration at 2 and 5 positions, 99% deuteration at 6 position, and 10% deuteration at 4 position when Pd/C was used alone.

IC ICM C07B059-00

C07C037-00; C07C039-06; C07C051-00; C07C057-30; C07C063-04;
C07C211-45; C07C209-00; C07D213-803; C07D213-74; C07B061-00;
C07M005-00

CC 27-16 (Heterocyclic Compounds (One Hetero Atom))
 Section cross-reference(s): 25

IT Deuteration

Deuteration catalysts

(method of deuteration of aromatic ring and/or heterocycle compds. using mixed metal catalyst such as palladium and platinum on carbon)

IT Deuteration

Deuteration catalysts

(regioselective; method of deuteration of aromatic ring and/or heterocycle compds. using mixed metal catalyst such as palladium and platinum on carbon)

TT 7439-88-5, Iridium, uses 7440-02-0, Nickel, uses 7440-05-3, Palladium, uses 7440-05-3D, Palladium, supported on carbon 7440-06-4, Platinum, uses 7440-06-4D, Platinum, supported on carbon 7440-16-6, Rhodium, uses 7440-18-8, Ruthenium, uses 7440-48-4, Cobalt, uses RL: CAT (Catalyst use); USES (Uses)

(method of deuteration of aromatic ring and/or heterocycle compds. using mixed metal catalyst such as palladium and platinum on carbon)

IT 1821-12-1P, 4-Phenylbutanoic acid 2438-05-3DP, deuterated derivative 7128-85-0P 7217-47-2DP, deuterated derivative 22527-01-1DP, deuterated derivative 22527-01-1P 66148-15-0P 87385-38-4DP, deuterated derivative

134860-14-3DP, Benzenebutanoic- α , α , β , β , . gamm

a., γ-d6 acid, deuterated derivative 358730-86-6P, Benzene-d5-butanoic-d6 acid 767627-97-4P, Benzene-d5-pentanoicd8 acid 861405-57-4DP, Benzenepentanoic-d8 acid, deuterated derivative 861405-58-5DP, Benzene-3,4,5-d3-pentanoic acid, deuterated derivative 861405-59-6DP, deuterated derivative 861405-60-9DP, Benzene-3,4,5-d3-butanoic acid, deuterated derivative 861405-61-0DP, deuterated derivative 861405-62-1P 861405-63-2P 861405-64-3DP, deuterated derivative 861405-65-4P 861405-66-5DP, deuterated derivative 861405-67-6DP, deuterated derivative 861405-68-7P 861405-69-8P 861405-70-1DP, deuterated derivative 861405-71-2P

861405-72-3P 861405-73-4DP, deuterated derivative 861405-74-5DP, deuterated derivative 861405-75-6P 861405-76-7DP, 3-Pyridine-6-d-

carboxylic acid, deuterated derivative

RL: SPN (Synthetic preparation); PREP (Preparation)
(method of deuteration of aromatic ring and/or heterocycle compds. using mixed metal catalyst such as palladium and platinum on carbon)

TT 7440-02-0, Nickel, uses 7440-05-3, Palladium, uses 7440-05-3D, Palladium, supported on carbon 7440-06-4, Platinum, uses 7440-06-4D, Platinum, supported on carbon 7440-16-6, Rhodium, uses 7440-18-8, Ruthenium, uses 7440-48-4, Cobalt, uses

```
RL: CAT (Catalyst use); USES (Uses)
```

(method of deuteration of aromatic ring and/or heterocycle compds. using mixed metal catalyst such as palladium and platinum on carbon)

RN 7440-02-0 CAPLUS

CN Nickel (CA INDEX NAME)

Νi

RN 7440-05-3 CAPLUS

CN Palladium (CA INDEX NAME)

Pd

RN 7440-05-3 CAPLUS

CN Palladium (CA INDEX NAME)

Pd

RN 7440-06-4 CAPLUS

CN Platinum (CA INDEX NAME)

Рt

RN 7440-06-4 CAPLUS

CN Platinum (CA INDEX NAME)

Рt

RN 7440-16-6 CAPLUS

CN Rhodium (CA INDEX NAME)

Rh

RN 7440-18-8 CAPLUS

CN Ruthenium (CA INDEX NAME)

Ru

RN 7440-48-4 CAPLUS CN Cobalt (CA INDEX NAME)

CO

IT 134860-14-3DP, Benzenebutanoic-α,α,β,β,.gamm
 a.,γ-d6 acid, deuterated derivative 358730-86-6P,
 Benzene-d5-butanoic-d6 acid 767627-97-4P, Benzene-d5-pentanoic d8 acid 861405-57-4DP, Benzenepentanoic-d8 acid, deuterated
 derivative 861405-58-5DP, Benzene-3,4,5-d3-pentanoic acid,
 deuterated derivative 861405-59-6DP, deuterated derivative
 861405-60-9DP, Benzene-3,4,5-d3-butanoic acid, deuterated derivative
 RL: SPN (Synthetic preparation); PREP (Preparation)
 (method of deuteration of aromatic ring and/or heterocycle compds. using
 mixed metal catalyst such as palladium and platinum on carbon)
 134860-14-3 CAPLUS
 Benzenebutanoic-α,α,β,β,γ,γ-d6 acid

(9CI) (CA INDEX NAME)

Ph- (CD2)3-CO2H

RN 358730-86-6 CAPLUS
CN Benzene-d5-butanoic-d6 acid (9CI) (CA INDEX NAME)

RN 767627-97-4 CAPLUS

CN Benzene-d5-pentanoic-d8 acid (CA INDEX NAME)

$$D \longrightarrow D \qquad (CD_2)_4 - CO_2H$$

RN 861405-57-4 CAPLUS

CN Benzenepentanoic-d8 acid (9CI) (CA INDEX NAME)

Ph- (CD2)4-CO2H

RN 861405-58-5 CAPLUS

CN Benzene-3,4,5-d3-pentanoic acid (9CI) (CA INDEX NAME)

RN 861405-59-6 CAPLUS

CN Benzene-3,4,5-d3-pentanoic- α , α , β , β , γ , γ ,. delta., δ -d8 acid (9CI) (CA INDEX NAME)

RN 861405-60-9 CAPLUS

CN Benzene-3,4,5-d3-butanoic acid (9CI) (CA INDEX NAME)

REFERENCE COUNT:

7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

10/539,188 L19 ANSWER 7 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:711257 CAPLUS Full-text DOCUMENT NUMBER: 141:379678 TITLE: Complete Replacement of H2 by D2 via Pd/C-Catalyzed H/D Exchange Reaction AUTHOR(S): Sajiki, Hironao; Kurita, Takanori; Esaki, Hiroyoshi; Aoki, Fumiyo; Maegawa, Tomohiro; Hirota, Kosaku CORPORATE SOURCE: Laboratory of Medicinal Chemistry, Gifu Pharmaceutical University, Gifu, 502-8585, Japan SOURCE: Organic Letters (2004), 6(20), 3521-3523 CODEN: ORLEF7; ISSN: 1523-7060 PUBLISHER: American Chemical Society DOCUMENT TYPE: Journal LANGUAGE: English OTHER SOURCE(S): CASREACT 141:379678 A general and in situ D2 gas generation method using 10% Pd/C-catalyzed H2-D2 exchange reaction in a H2-D2O system has been developed. H2 gas sealed in a reaction flask was efficiently converted into nearly pure D2 gas, which can be used for the reductive deuteration of substrates possessing reducible functionalities within the mol. CC 25-17 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds) IT Deuteration Deuteration catalysts Hydrogenation Hydrogenation catalysts (palladium-catalyzed reductive deuteration using hydrogen and deuterium IT **7440-05-3**, Palladium, uses RL: CAT (Catalyst use); USES (Uses) (palladium-catalyzed reductive deuteration using hydrogen and deuterium 100-46-9P, Benzylamine, preparation IT 123-25-1P, Diethyl succinate 4551-39-7P, 2-Deuteriobenzoic acid 4551-62-6P, 4-Deuteriobenzoic acid **782486-40-2P**, Benzenepropanoic- α , β -d2 acid-d 782486-42-4P, Benzene-3-d-propanoic acid 782486-43-5P, Benzene-3-d-ethanol RL: SPN (Synthetic preparation); PREP (Preparation) (palladium-catalyzed reductive deuteration using hydrogen and deuterium oxide)

IT 7440-05-3, Palladium, uses

RL: CAT (Catalyst use); USES (Uses)

(palladium-catalyzed reductive deuteration using hydrogen and deuterium oxide)

7440-05-3 CAPLUS RN

CN Palladium (CA INDEX NAME)

Pd

- **782486-40-2P**, Benzenepropanoic- α , β -d2 acid-d IT 782486-42-4P, Benzene-3-d-propanoic acid
 - RL: SPN (Synthetic preparation); PREP (Preparation)

(palladium-catalyzed reductive deuteration using hydrogen and deuterium oxide)

RN782486-40-2 CAPLUS

CN Benzenepropanoic- α , β -d2 acid-d (9CI) (CA INDEX NAME) Ph-CH-CH-C-0-D

782486-42-4 CAPLUS RN

CN Benzene-3-d-propanoic acid (9CI) (CA INDEX NAME)

.CH2-CH2-CO2H

REFERENCE COUNT:

37 THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 8 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER:

2004:589514 CAPLUS Full-text

DOCUMENT NUMBER:

141:139883

TITLE:

Method of catalytic deuteration of carbonyl compounds

or secondary alcohols by heavy water

INVENTOR(S):

Ito, Nobuhiro; Maesawa, Tsuneaki; Muto, Kazushige;

Hirota, Kosaku; Sajiki, Hironao

PATENT ASSIGNEE(S):

Wako Pure Chemical Industries, Ltd., Japan

SOURCE:

PCT Int. Appl., 42 pp. CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PAT	PATENT NO.				KIND DATE				APPLICATION NO.									
WO	WO 2004060831				A1	20040722			WO 2003-JP14182									
	W:	ΑE,	AG,	AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	ΒZ,	CA,	CH,	CN,	
		CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	
													KR,					
													MZ,					
													SL,					
													ZM,					
	RW:	GH,	GM,	ΚE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	ΑZ,	BY,	
		KG,	KZ,	MD,	RU,	ТJ,	TM,	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	
													RO,					
		BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	TG	
CA					A 1		2004	0722	CA 2003-2511885						20031107			
AU	AU 2003277596				A1		2004	0729	AU 2003-277596						20031107			
EP	EP 1577280			A 1		2005	0921	EP 2003-814536						20031107				
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	ΙΤ,	LI,	LU,	NL,	SE,	MC,	PT,	
		ΙE,	SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL,	TR,	BG,	CZ,	EE,	HU,	SK		
CN	1732	135			Α	:	2006	0208	(CN 2	003-	3010	7483		20	0031	L07	
US	US 2006116535				A1	20060601								20050616				
IN	2005	KN01	449		Α		2007	0720	:	IN 20	005-1	KN14	49		20	0050	726	
RIORITY	ORITY APPLN. INFO.:									JP 2002-378932				7	A 20021227			

WO 2003-JP14182 W 20031107

OTHER SOURCE(S): CASREACT 141:139883; MARPAT 141:139883 Described is a method of deuterating a carbonyl or secondary alc. compound AB represented by the general formula R1-X-R2 (I) (wherein R1 = alkyl optionally possessing a CH:CH or C.tplbond.C bond, aralkyl; R2 = alkyl optionally possessing a CH:CH or C.tplbond.C bond, aryl, aralkyl, alkoxy, aryloxy, hydroxy; X carbonyl, hydroxymethylene), which comprises reacting the compound represented by the general formula I with a deuterium source, in particular D20, in the presence of a catalyst selected among activated palladium, platinum, rhodium, ruthenium, nickel, and cobalt catalysts. By the method, deuteration, which has been conducted under severe conditions, can be conducted under neutral conditions. Even when the compound contains an unsatd. bond, it can be deuterated without reducing the unsatd. bond. Not only hydrogens near the carbonyl or hydroxymethylene group but also those remotely situated from these groups are selectively deuterated without deuterating the carbon-carbon double or triple bonds. Thus, 500 mg tricyclo[5.2.1.02'6]decan-8-ol and 100 mg Pd-C were suspended in 17 mL D2O, purged with H, and heated at 180° for 24 h in an oil bath to give tricyclo[5.2.1.02'6]decan-8-ol deuterated by 96% at 8-position and 88% at other positions. IC ICM C07B059-00 ICS C07C029-00; C07C031-02; C07C035-08; C07C035-29; C07C035-37; C07C045-00; C07C049-04; C07C049-08; C07C049-433; C07C049-453; C07C051-00; C07C053-10; C07C053-124; C07C057-04; C07M005-00 CC 21-2 (General Organic Chemistry) ITDeuteration Deuteration catalysts (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) 7440-02-0, Raney nickel, uses IT RL: CAT (Catalyst use); USES (Uses) (catalysts; catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) 7440-05-3, Palladium, uses 7440-05-3D, Palladium, ΙT supported on carbon 7440-06-4, Platinum, uses 7440-06-4D , Platinum, supported on carbon 7440-16-6, Rhodium, uses 7440-16-6D, Rhodium, supported on alumina 7440-16-6D, Rhodium, supported on carbon 7440-18-8, Ruthenium, uses 7440-18-8D, Ruthenium, supported on carbon 7440-48-4, Cobalt, uses RL: CAT (Catalyst use); USES (Uses) (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) IT 79-31-2DP, Isobutyric acid, deuterated 108-93-0DP, Cyclohexanol, deuterated 666-52-4P, 2-Propanone-1,1,1,3,3,3-d6 13380-89-7DP, Tricyclo[5.2.1.02,6]decan-8-ol, deuterated 14044-94-1P 18153-61-2DP, Bicyclo[2.2.1]heptan-2-one-3,3-d2, deuterated 21273-02-9DP, Cyclohexan-1-d-ol, deuterated 51209-49-5P, Cyclohexanone-d10 53481-06-4P 55935-44-9P 63870-91-7DP, Norbornenol, deuterated 64118-21-4P 91468-78-9DP, Bicyclo[2.2.1]heptan-2-d-2-ol, deuterated 350820-09-6P 725242-18-2P, 4-Heptanone-d14 725242-19-3DP, deuterated 725242-21-7DP, deuterated **725242-22-8P** , 2-Heptanone-d14 725242-23-9P, 3-Heptanone-d14 725242-24-0P, 2-Heptanol-d15 725242-25-1P, 4-Heptanol-d15 725242-26-2DP, deuterated 725242-27-3DP , deuterated 725242-28-4DP, deuterated 725242-29-5DP,

deuterated 725242-29-5P 725242-30-8P

```
725242-31-9P
                    725242-32-0DP, deuterated
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (catalytic deuteration of carbonyl compds. or secondary alc. compds.
        with heavy water in presence of palladium, platinum, rhodium,
        ruthenium, or nickel)
ΙT
     7440-02-0, Raney nickel, uses
     RL: CAT (Catalyst use); USES (Uses)
        (catalysts; catalytic deuteration of carbonyl compds. or secondary alc.
        compds. with heavy water in presence of palladium, platinum, rhodium,
        ruthenium, or nickel)
     7440-02-0 CAPLUS
RN
CN
     Nickel (CA INDEX NAME)
 Νi
ΙT
     7440-05-3, Palladium, uses 7440-05-3D, Palladium,
     supported on carbon 7440-06-4, Platinum, uses 7440-06-4D
     , Platinum, supported on carbon 7440-16-6, Rhodium, uses
     7440-16-6D, Rhodium, supported on alumina 7440-18-8,
     Ruthenium, uses 7440-18-8D, Ruthenium, supported on carbon
     7440-48-4, Cobalt, uses
     RL: CAT (Catalyst use); USES (Uses)
        (catalytic deuteration of carbonyl compds. or secondary alc. compds.
        with heavy water in presence of palladium, platinum, rhodium,
        ruthenium, or nickel)
     7440-05-3 CAPLUS
RN
CN
     Palladium (CA INDEX NAME)
 Pd
     7440-05-3 CAPLUS
RN
CN
     Palladium (CA INDEX NAME)
 Pd
     7440-06-4 CAPLUS
RN
CN
     Platinum (CA INDEX NAME)
Ρt
```

RN 7440-06-4 CAPLUS CN Platinum (CA INDEX NAME) Ρt

RN 7440-16-6 CAPLUS CN Rhodium (CA INDEX NAME)

Rh

RN 7440-16-6 CAPLUS CN Rhodium (CA INDEX NAME)

Rh

RN 7440-18-8 CAPLUS CN Ruthenium (CA INDEX NAME)

Ru

RN 7440-18-8 CAPLUS
CN Ruthenium (CA INDEX NAME)

Ru

RN 7440-48-4 CAPLUS CN Cobalt (CA INDEX NAME)

Со

IT 666-52-4P, 2-Propanone-1,1,1,3,3,3-d6 14044-94-1P 21273-02-9DP, Cyclohexan-1-d-ol, deuterated 53481-06-4P 55935-44-9P 64118-21-4P 91468-78-9DP, Bicyclo[2.2.1]heptan-2-d-2-ol, deuterated 350820-09-6P 725242-18-2P, 4-Heptanone-dl4 725242-22-8P, 2-Heptanone-dl4 725242-23-9P, 3-Heptanone-dl4 725242-24-0P, 2-Heptanol-dl5 725242-25-1P, 4-Heptanol-dl5 725242-26-2DP, deuterated 725242-27-3DP

, deuterated 725242-28-4DP, deuterated 725242-29-5DP,
deuterated 725242-29-5P 725242-30-8P
725242-31-9P

RL: SPN (Synthetic preparation); PREP (Preparation)

(catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel)

RN 666-52-4 CAPLUS

CN 2-Propanone-1,1,1,3,3,3-d6 (9CI) (CA INDEX NAME)

D₃C_C_CD₃

RN 14044-94-1 CAPLUS

CN Acetic-d3 acid, sodium salt (8CI, 9CI) (CA INDEX NAME)

Na Na

RN 21273-02-9 CAPLUS CN Cyclohexan-1-d-ol (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

RN 53481-06-4 CAPLUS
CN Acetic-d acid, sodium salt (9CI) (CA INDEX NAME)

о || но_ с_ сн₂ _ D

Na Na

RN 55935-44-9 CAPLUS CN 2-Propenoic-3,3-d2 acid, 2-(methyl-d3)- (CA INDEX NAME)

RN 64118-21-4 CAPLUS CN Acetic-d2 acid, sodium salt (9CI) (CA INDEX NAME)

D_CH_CO2H

Na Na

RN 91468-78-9 CAPLUS CN Bicyclo[2.2.1]heptan-2-d-2-ol (9CI) (CA INDEX NAME)

RN 350820-09-6 CAPLUS CN 2-Butanone-1,1,1,3,3,4,4,4-d8 (9CI) (CA INDEX NAME)

RN 725242-18-2 CAPLUS CN 4-Heptanone-1,1,1,2,2,3,3,5,5,6,6,7,7,7-d14 (9CI) (CA INDEX NAME)

RN 725242-22-8 CAPLUS CN 2-Heptanone-1,1,1,3,3,4,4,5,5,6,6,7,7,7-d14 (9CI) (CA INDEX NAME)

RN 725242-23-9 CAPLUS

CN 3-Heptanone-1,1,1,2,2,4,4,5,5,6,6,7,7,7-d14 (9CI) (CA INDEX NAME)

RN 725242-24-0 CAPLUS

CN 2-Heptan-1,1,1,2,3,3,4,4,5,5,6,6,7,7,7-d15-ol (9CI) (CA INDEX NAME)

RN 725242-25-1 CAPLUS

CN 4-Heptan-1,1,1,2,2,3,3,4,5,5,6,6,7,7,7-d15-ol (9CI) (CA INDEX NAME)

RN 725242-26-2 CAPLUS

CN 4,7-Methano-1H-inden-2,3-d2-5-ol, 3a,4,5,6,7,7a-hexahydro- (9CI) (CA INDEX NAME)

RN 725242-27-3 CAPLUS

CN 4,7-Methano-1H-inden-3-d-5-ol, 3a,4,5,6,7,7a-hexahydro- (9CI) (CA INDEX NAME)

RN 725242-28-4 CAPLUS

CN 4,7-Methano-1H-inden-2-d-5-ol, 3a,4,5,6,7,7a-hexahydro- (9CI) (CA INDEX NAME)

RN 725242-29-5 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-5-d- (9CI) (CA INDEX NAME)

RN 725242-29-5 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-5-d- (9CI) (CA INDEX NAME)

RN 725242-30-8 CAPLUS

CN 4,7-Methano-1H-inden-1,1,2,3,6,7,7-d7-5-ol, octahydro-2,3,3a,4,5,6,7,7a-d8-(9CI) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

RN 725242-31-9 CAPLUS

CN 2-Propenoic-3,3-d2 acid, 2-(methyl-d3)-, sodium salt (9CI) (CA INDEX NAME)

Na

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 9 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

10/539,188 ACCESSION NUMBER: 2004:453150 CAPLUS Full-text DOCUMENT NUMBER: 141:23545 TITLE: Method for deuteration or tritiation of heterocyclic INVENTOR(S): Ito, Nobuhiro; Maesawa, Tsuneaki; Muto, Kazushige; Hirota, Kosaku; Sajiki, Hironao PATENT ASSIGNEE(S): Wako Pure Chemical Industries, Ltd., Japan SOURCE: PCT Int. Appl., 45 pp. CODEN: PIXXD2 DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE ----_____

```
WO 2004046066
                                           WO 2003-JP14181
                         A1
                                20040603
                                                                   20031107
        W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
            CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
            GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
            LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM,
            PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN,
            TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
        RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
            BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE,
            ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK,
            TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
    CA 2506010
                         A1
                               20040603
                                         CA 2003-2506010
                                                                   20031107
    AU 2003277595
                         A1
                               20040615
                                           AU 2003-277595
                                                                   20031107
    EP 1561741
                         A1
                               20050810
                                           EP 2003-811499
                                                                   20031107
           AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
    CN 1714060
                               20051228
                         Α
                                           CN 2003-80103924
                                                                   20031107
    US 2006025596
                         A1
                               20060202
                                           US 2005-534344
                                                                   20050509
    IN 2005KN01145
                         Α
                               20061110
                                           IN 2005-KN1145
                                                                   20050615
PRIORITY APPLN. INFO.:
                                           JP 2002-331594
                                                               A 20021115
                                           WO 2003-JP14181
                                                               W 20031107
```

AB A method for deuteration or tritiation of a heterocyclic ring comprises allowing a heterocyclic compound to be present under a sealing and refluxing condition in a deuterated or tritiated solvent (e.g., D2O) in the presence of an activated catalyst selected from among a palladium catalyst, a platinum catalyst, a rhodium catalyst, a ruthenium catalyst, a nickel catalyst and a cobalt catalyst. The method allows a deuteration or tritiation temperature to be kept at a temperature higher than the boiling temperature of the solvent, which results in the replacement of a hydrogen atom in a heterocyclic ring of a heterocyclic compound with very good efficiency. Further, the method can be widely used for the deuteration or tritiation of various types of heterocyclic compds. in a com. process.

```
IC ICM C07B059-00
```

ICS C07D209-08; C07D209-20; C07D213-06; C07D231-12; C07D233-58; C07D235-08; C07D239-47; C07D239-54; C07D471-04; C07D473-30; C07D473-34; C07H019-067; C07H019-167; C07M005-00

CC 28-16 (Heterocyclic Compounds (More Than One Hetero Atom)) Section cross-reference(s): 26, 33, 34

IT Deuteration

Tritiation

(method for deuteration or tritiation of heterocyclic compds.)

IT Deuteration catalysts

Tritiation catalysts

(palladium and platinum, in method for deuteration or tritiation of

```
heterocyclic compds.)
ΙT
     7440-02-0, Nickel, uses 7440-05-3, Palladium, uses
     7440-06-4, Platinum, uses 7440-16-6, Rhodium, uses
     7440-18-8, Ruthenium, uses
                                  7440-44-0, Carbon, uses
     7440-48-4, Cobalt, uses
     RL: CAT (Catalyst use); USES (Uses)
        (method for deuteration or tritiation of heterocyclic compds.)
IT
                  6745-43-3P, 1H-Imidazole-2,4,5-d3
     4166-68-1P
                                                      22194-79-2P
     24897-52-7P, 2,4(1H,3H)-Pyrimidinedione-5,6-d2 40632-21-1P,
     Uridine-5,6-d2
                     62595-11-3P, L-Tryptophan-2,4,5,6,7-d5
     82845-88-3P, Adenosine-2,8-d2 96412-41-8P, Guanosine-8-d
                    130317-91-8P
                                 200496-79-3P
     106391-24-6P
                                                350818-65-4P
                    697807-00-4P, 1H-Purin-2,8-d2-6-amine 697807-01-5P
     697806-99-8P
                                       697807-03-7P
     , Inosine-2,8-d2
                        697807-02-6P
                                                      697807-04-8P
     697807-05-9P
                    697807-06-0P
                                   697807-07-1P
     RL: IMF (Industrial manufacture); SPN (Synthetic
     preparation); PREP (Preparation)
        (method for deuteration or tritiation of heterocyclic compds.)
     7440-02-0, Nickel, uses 7440-05-3, Palladium, uses
IT
     7440-06-4, Platinum, uses 7440-16-6, Rhodium, uses
     7440-18-8, Ruthenium, uses 7440-48-4, Cobalt, uses
     RL: CAT (Catalyst use); USES (Uses)
        (method for deuteration or tritiation of heterocyclic compds.)
RN
     7440-02-0 CAPLUS
CN
    Nickel (CA INDEX NAME)
Νi
     7440-05-3 CAPLUS
RN
    Palladium (CA INDEX NAME)
CN
Pd
RN
     7440-06-4 CAPLUS
CN
     Platinum (CA INDEX NAME)
Pt
RN
    7440-16-6 CAPLUS
CN
    Rhodium (CA INDEX NAME)
```

Rh

RN 7440-18-8 CAPLUS

CN Ruthenium (CA INDEX NAME)

Ru

RN 7440-48-4 CAPLUS CN Cobalt (CA INDEX NAME)

Со

IT 40632-21-1P, Uridine-5,6-d2 82845-88-3P,
 Adenosine-2,8-d2 96412-41-8P, Guanosine-8-d 697807-01-5P
 , Inosine-2,8-d2
 RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP (Preparation)
 (method for deuteration or tritiation of heterocyclic compds.)
RN 40632-21-1 CAPLUS
CN Uridine-5,6-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

RN 82845-88-3 CAPLUS
CN Adenosine-2,8-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

RN 96412-41-8 CAPLUS
CN Guanosine-8-d (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

697807-01-5 CAPLUS

CN Inosine-2,8-d2 (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 10 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:231394 CAPLUS <u>Full-text</u>

DOCUMENT NUMBER:

140:406566

TITLE:

RN

Palladium-catalyzed H-D exchange reaction under

hydrothermal condition

AUTHOR(S):

Matsubara, Seijiro; Yokota, Yutaka; Oshima, Koichiro Department of Material Chemistry, Graduate School of

CORPORATE SOURCE:

Engineering, Kyoto University, Kyoto, 615-8510, Japan

SOURCE: Chemistry Letters (2004), 33(3), 294-295

CODEN: CMLTAG; ISSN: 0366-7022

PUBLISHER: Chemical Society of Japan DOCUMENT TYPE: Journal

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 140:406566

AB Alkenes and alkanes were converted into fully deuterium labeled ones by treatment with palladium on charcoal and deuterium oxide under hydrothermal condition. This simple method to get fully deuterium labeled compds. is easy to apply to various types of organic compds.

CC 24-6 (Alicyclic Compounds)

IT Deuteration

Deuteration catalysts

 $\begin{tabular}{ll} (palladium-catalyzed $\mbox{ H-$D}$ exchange reaction under hydrothermal conditions) \end{tabular}$

IT **7440-05-3**, Palladium, uses

RL: CAT (Catalyst use); USES (Uses)

(palladium-catalyzed H-D exchange reaction under hydrothermal conditions) IT 10249-89-5P, Cyclooctene-d14 1486-01-7P 16450-78-5P, Cyclododecane-d24 688320-41-4P 36340-20-2P, Pentadecane-d32 688320-42-5P, Cyclopentadecane-d30 688320-43-6P, Cyclooctanone-d14 688320-44-7P, Cyclodecanone-d18 688320-45-8P, Cyclododecanone-d22 688320-46-9P,

Cyclopentadecanone-d28 688320-48-1P 688320-49-2P RL: SPN (Synthetic preparation); PREP (Preparation)

(palladium-catalyzed H-D exchange reaction under hydrothermal conditions)

7440-05-3, Palladium, uses IT

RL: CAT (Catalyst use); USES (Uses)

(palladium-catalyzed H-D exchange reaction under hydrothermal conditions)

7440-05-3 CAPLUS RN

CN Palladium (CA INDEX NAME)

Pd

ΙT 688320-48-1P

RL: SPN (Synthetic preparation); PREP (Preparation)

(palladium-catalyzed H-D exchange reaction under hydrothermal conditions)

688320-48-1 CAPLUS RN ·

2-Dodecanone-1,1,1,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12,12-d24 CN (CA INDEX NAME)

D3C-(CD2)9-C-CD3

REFERENCE COUNT: 23 THERE ARE 23 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

CAPLUS COPYRIGHT 2008 ACS on STN. L19 ANSWER 11 OF 22 ACCESSION NUMBER:

2003:991461 CAPLUS Full-text

DOCUMENT NUMBER:

140:41620

TITLE:

Process for deuteration of inert methylene

INVENTOR(S): Hirota, Kosaku; Sajiki, Hironao

PATENT ASSIGNEE(S):

Wako Pure Chemical Industries, Ltd., Japan

SOURCE:

PCT Int. Appl., 27 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.		DATE	
WO 2003104166 W: JP, US	A1	20031218	WO 2002-JP11785		20021112	
US 2005177015	A1	20050811	US 2004-516638		20041202	
US 7126023	B2	20061024				
PRIORITY APPLN. INFO.:			JP 2002-166224	Α	20020606	
			WO 2002-JP11785	W	20021112	

10/539,188 OTHER SOURCE(S): MARPAT 140:41620 AB The invention relates to a process for deuteration of inert alkanes with activated palladium-carbon, specifically, a process for deuterating a compound having either a Me group or an alkylene group having two or more carbon atoms in a state directly bonded to an optionally substituted aromatic ring through replacement of one or more hydrogen atoms of the Me group or one or more of the benzylic and other hydrogen atoms of the alkylene group by deuterium, characterized in that the above compound is subjected to refluxing in a closed system in the presence of activated palladium-carbon in a state dissolved in a deuterated solvent. IC ICM C07B059-00 ICS C07M005-00 21-2 (General Organic Chemistry) CC ΙT Deuteration Deuteration catalysts (process for deuteration of inert methylene using activated palladium-carbon catalyst) **7440-05-3**, Palladium, uses ΙT RL: CAT (Catalyst use); USES (Uses) (carbon-supported; process for deuteration of inert methylene using activated palladium-carbon catalyst) IT 1124-18-1P 14202-49-4P 38729-11-2P 65087-58-3P 94367-56-3P 117637-87-3P, Benzenepropan- β , β , γ , γ -d4-ol 156310-21-3P **634897-72-6P** 634897-76-0P 634897-78-2P 634897-80-6P 634897-84-0P 634897-86-2P, 634897-82-8P Benzenepropanoic-d4 acid-d 634897-88-4P, Benzene-d5-butanoic-d6 acid-d 634897-90-8P 634897-92-0P 634897-99-7P 634898-02-5P, Benzenebutan- β , β , γ , γ , δ , δ -634898-04-7P, Benzenepentan- β , β , γ , γ , δ , $\delta \varepsilon \varepsilon = d\theta - o1$ 634898-08-1P 634898-10-5P 634898-14-9P 634898-16-1P 634898-20-7P 634898-23-0P 634898-25-2P

634898-29-6P 634898-34-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(process for deuteration of inert methylene using activated palladium-carbon catalyst)

IT **7440-05-3**, Palladium, uses

RL: CAT (Catalyst use); USES (Uses)

(carbon-supported; process for deuteration of inert methylene using activated palladium-carbon catalyst)

RN 7440-05-3 CAPLUS

CN Palladium (CA INDEX NAME)

Pd

634897-72-6P 634897-86-2P, Benzenepropanoic-d4 acid-d IT 634897-88-4P, Benzene-d5-butanoic-d6 acid-d 634897-90-8P 634897-92-0P 634897-99-7P RL: SPN (Synthetic preparation); PREP (Preparation)

(process for deuteration of inert methylene using activated

palladium-carbon catalyst)

RN 634897-72-6 CAPLUS

Benzene-d5-pentanoic-d8 acid, sodium salt (9CI) (CA INDEX NAME) CN

$$D \longrightarrow D \qquad (CD_2)_4 - CO_2H$$

Na

RN 634897-86-2 CAPLUS

CN Benzenepropanoic-d4 acid-d (9CI) (CA INDEX NAME)

RN 634897-88-4 CAPLUS

CN Benzene-d5-butanoic-d6 acid-d (9CI) (CA INDEX NAME)

$$D \longrightarrow D \longrightarrow D \longrightarrow D$$

$$D \longrightarrow D$$

$$D \longrightarrow D$$

RN 634897-90-8 CAPLUS

CN Benzene-d5-pentanoic- β , β , γ , γ , δ , δ -d6 acid-d (9CI) (CA INDEX NAME)

$$\begin{array}{c} D \\ D \\ D \\ \end{array} \begin{array}{c} (CD_2)_3 - CH_2 - \overset{O}{C} - O - D \\ \end{array}$$

RN 634897-92-0 CAPLUS

CN Benzene-d5-hexanoic- β , β , γ , γ , δ , δ , epsilon ., ϵ -d8 acid-d (9CI) (CA INDEX NAME)

$$\begin{array}{c} D \\ D \\ D \\ \end{array} \begin{array}{c} (CD_2) \ 4 - CH_2 - C - O - D \\ \end{array}$$

RN634897-99-7 CAPLUS

Benzenepentanoic- β , β , γ , γ , δ , δ -d6 acid, CN methyl ester (9CI) (CA INDEX NAME)

 $-CH_2-(CD_2)_3-Ph$

REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 12 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2003:474175 CAPLUS Full-text

DOCUMENT NUMBER:

139:395665

TITLE: Combining microwave-enhanced deuteration reactions

with parallel synthesis procedures

AUTHOR (S): Chappelle, Michael R.; Harding, John R.; Kent, Barry

B.; Jones, John R.; Lu, Shui-Yu; Morgan, Alan D.

CORPORATE SOURCE: Amersham Biosciences, The Maynard Centre, Cardiff,

CF14 7YT, UK

Journal of Labelled Compounds & Radiopharmaceuticals SOURCE:

(2003), 46(6), 567-574

CODEN: JLCRD4; ISSN: 0362-4803

PUBLISHER: John Wiley & Sons Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 139:395665

The development of combined microwave-enhanced/parallel synthesis procedures AΒ and their application to the deuteration of organic compds. via examples of solid-state hydrogenation is reported. Other labeling procedures, such as solution state catalytic dehalogenations, hydrogenations as well as hydrogen isotope exchange reactions also benefit from the combined technol.

CC 25-17 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)

Debromination Dechlorination Dehalogenation

Deuteration catalysts

Microwave

Solid phase synthesis

(combining microwave-enhanced deuteration with parallel synthesis procedures)

IT Deuteration

Hydrogenation catalysts

(solid-state deuteration; combining microwave-enhanced deuteration with parallel synthesis procedures)

IT 3375-31-3 10049-07-7, Rhodium chloride (RhCl3)

```
RL: CAT (Catalyst use); USES (Uses)
        (combining microwave-enhanced deuteration with parallel synthesis
        procedures)
TT
     10473-16-2P
                   16089-48-8P, 3-Phenyl-2-Propenoic acid potassium salt
     36568-19-1P, 3-(4-Chlorophenyl)-2-Propenoic acid potassium salt
     625383-67-7P, 3-(4-Bromophenyl)-2-propenoic acid potassium salt
     625383-68-8P, 4-Bromobenzenepropanoic-\alpha, \beta-d2 acid potassium
     salt 625383-69-9P, 3-(Phenyl-4-d)-2-Propenoic acid potassium
            625383-70-2P, 3-(3-Bromophenyl)-2-propenoic acid potassium salt
     625383-72-4P 625383-73-5P, 3-(Phenyl-3-d)-2-Propenoic acid
                      625383-74-6P, 1-Bromo-4-(ethyl-1,2-d2)benzene
     potassium salt
     625383-75-7P, 3-(4-Fluorophenyl)-2-Propenoic acid potassium salt
     625383-76-8P, 3-(2-Chlorophenyl)-2-Propenoic acid potassium salt
     625383-77-9P, 3-(3-Chlorophenyl)-2-Propenoic acid potassium salt
                    625383-80-4P, 4-Fluorobenzenepropanoic-
     \alpha, \beta-d2 acid potassium salt
                                   625383-82-6P, 2-
     Chlorobenzenepropanoic-\alpha, \beta-d2 acid potassium salt
     625383-84-8P, 3-Chlorobenzenepropanoic-\alpha, \beta-d2 acid potassium
            625383-86-0P, 4-Chlorobenzenepropanoic-\alpha, \beta-d2 acid
     potassium salt
                      625383-88-2P, 1-Bromo-4-(ethenyl-1,2-d2)benzene
     625383-90-6P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (combining microwave-enhanced deuteration with parallel synthesis
        procedures)
ΙT
     3375-31-3 10049-07-7, Rhodium chloride (RhCl3)
     RL: CAT (Catalyst use); USES (Uses)
        (combining microwave-enhanced deuteration with parallel synthesis
        procedures)
     3375-31-3 CAPLUS
RN
     Acetic acid, palladium(2+) salt (2:1) (CA INDEX NAME)
CN
 ●1/2 Pd(II)
RN
     10049-07-7 CAPLUS
CN
     Rhodium chloride (RhCl3) (CA INDEX NAME)
    Cl
 C1-Rh-C1
     625383-69-9P, 3-(Phenyl-4-d)-2-Propenoic acid potassium salt
     625383-73-5P, 3-(Phenyl-3-d)-2-Propenoic acid potassium salt
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (combining microwave-enhanced deuteration with parallel synthesis
        procedures)
RN
     625383-69-9 CAPLUS
CN
     2-Propenoic acid, 3-(phenyl-4-d)-, potassium salt (9CI) (CA INDEX NAME)
```

K

RN625383-73-5 CAPLUS

2-Propenoic acid, 3-(phenyl-3-d)-, potassium salt (9CI) (CA INDEX NAME) CN

RN625383-78-0 CAPLUS

CN Benzenepropanoic- α , β -d2 acid, potassium salt (9CI) (CA INDEX NAME)

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 13 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2002:483457 CAPLUS Full-text

DOCUMENT NUMBER:

138:4188

TITLE:

Development of combined microwave-enhanced labelling procedures for maximizing deuterium incorporation

AUTHOR(S):

Chapelle, Michael R.; Kent, Barry B.; Jones, John R.; Lu, Shui-Yu; Morgan, Alan D.

CORPORATE SOURCE:

Amersham Plc, Cardiff Laboratories, Cardiff, CF14 7YT,

Tetrahedron Letters (2002), 43(29), 5117-5118 CODEN: TELEAY; ISSN: 0040-4039

SOURCE: PUBLISHER:

Elsevier Science Ltd.

DOCUMENT TYPE:

Journal

LANGUAGE:

English

OTHER SOURCE(S):

CASREACT 138:4188

- AB Combined hydrogenation/aromatic dehalogenation under microwave-enhanced conditions provides a rapid route to deuterium labeled compds. with enhanced isotopic incorporation.
- CC 21-2 (General Organic Chemistry)
- IT Debromination

Debromination catalysts

Dechlorination

Dechlorination catalysts

Deuteration

Deuteration catalysts

Hydrogenation

Hydrogenation catalysts

Microwave

(preparation of deuterium-labeled compds. with enhanced isotopic incorporation via rhodium- and palladium-catalyzed microwave-induced combined hydrogenation/aromatic dehalogenation procedure)

IT 3375-31-3, Palladium diacetate 10049-07-7, Rhodium chloride (RhCl3)

RL: CAT (Catalyst use); USES (Uses)

(microwave-induced rhodium- and palladium-catalyzed combined hydrogenation/aromatic dehalogenation routes for preparation of deuterium-labeled compds. with enhanced isotopic incorporation)

IT 99532-30-6P 477284-15-4P 477284-16-5P,

Benzene-4-d-propanoic- α , β -d2 acid

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of deuterium-labeled compds. with enhanced isotopic incorporation via combined microwave-induced hydrogenation/aromatic dehalogenation procedure)

- IT 477284-17-6P, Benzene-3-d-propanoic- α , β -d2 acid
 - RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of deuterium-labeled compds. with enhanced isotopic incorporation via microwave-induced hydrogenation/aromatic dehalogenation procedure)

RL: CAT (Catalyst use); USES (Uses)

(microwave-induced rhodium- and palladium-catalyzed combined hydrogenation/aromatic dehalogenation routes for preparation of deuterium-labeled compds. with enhanced isotopic incorporation)

RN 3375-31-3 CAPLUS

CN Acetic acid, palladium(2+) salt (2:1) (CA INDEX NAME)

но— с— сн3

●1/2 Pd(II)

RN 10049-07-7 CAPLUS

CN Rhodium chloride (RhCl3) (CA INDEX NAME)

C1 C1—Rh—C1 IT 99532-30-6P 477284-16-5P, Benzene-4-d-propanoic-

 α, β -d2 acid

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of deuterium-labeled compds. with enhanced isotopic incorporation via combined microwave-induced hydrogenation/aromatic dehalogenation procedure)

RN 99532-30-6 CAPLUS

CN 2-Propenoic acid, 3-(phenyl-4-d)- (9CI) (CA INDEX NAME)

RN 477284-16-5 CAPLUS

CN Benzene-4-d-propanoic- α , β -d2 acid (9CI) (CA INDEX NAME)

IT 477284-17-6P, Benzene-3-d-propanoic- α , β -d2 acid

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of deuterium-labeled compds. with enhanced isotopic incorporation via microwave-induced hydrogenation/aromatic dehalogenation procedure)

RN 477284-17-6 CAPLUS

CN Benzene-3-d-propanoic- α , β -d2 acid (9CI) (CA INDEX NAME)

REFERENCE COUNT: 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 14 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2000:550049 CAPLUS Full-text

DOCUMENT NUMBER:

133:322037

TITLE:

Deuteration of estrogens using Pd/C as a catalyst

AUTHOR(S): Kiuru, Paula; Wahala, Kristiina

CORPORATE SOURCE:

Department of Chemistry, Organic Chemistry Laboratory,

SOURCE:

University of Helsinki, FIN-00014, Finland

Synthesis and Applications of Isotopically Labelled Compounds 1997, Proceedings of the International

Symposium, 6th, Philadelphia, PA, United States, Sept.

14-18, 1997 (1998), Meeting Date 1997, 475-477. Editor(s): Heys, J. Richard; Melillo, David G. John

Wiley & Sons Ltd.: Chichester, UK.

CODEN: 69AGFQ

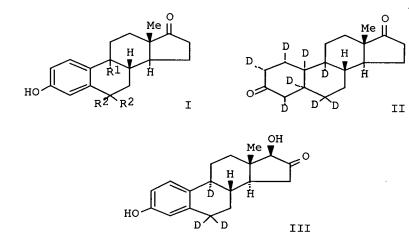
DOCUMENT TYPE:

LANGUAGE:

Conference

GΙ

English



The reduction of estrone (I; R1 = R2 = H) using D2 on Pd/C gives $1\alpha, 2\alpha, 4\alpha, 5\alpha, 6, 6, 9\alpha, 10\alpha$ -[2H3]estrane- 3,17-dione (II), the configuration of deuteriums been established by NMR. Pd/C catalyzes the H-D exchange also at the benzylic positions of estrogens. 6,6,9-[2H3]estrone (I; R1 = R2 = D) and 6,6,9-[2H3]-16- ketoestradiol (III) were synthesized in high isotopic purity.

CC 32-3 (Steroids)

IT Absolute configuration

Deuteration

Deuteration catalysts

(stereoselective deuteration of estrogens using palladium/carbon as a catalyst)

IT 7440-05-3D, Palladium, on carbon, uses

RL: CAT (Catalyst use); USES (Uses)

(stereoselective deuteration of estrogens using palladium/carbon as a catalyst)

IT 50888-33-0P, 6,6,9-[2H3]Estrone 303128-11-2P,

 1α , 5α , 6, 6, 9α , 10α -[2H6] Estrane-3, 17-dione

303128-12-3P, 6,6,9-[2H3]-16-Ketoestradiol

RL: SPN (Synthetic preparation); PREP (Preparation)

(stereoselective deuteration of estrogens using palladium/carbon as a catalyst)

IT 7440-05-3D, Palladium, on carbon, uses

RL: CAT (Catalyst use); USES (Uses)

(stereoselective deuteration of estrogens using palladium/carbon as a catalyst)

RN 7440-05-3 CAPLUS

CN Palladium (CA INDEX NAME)

Pd

IT **303128-12-3P**, 6,6,9-[2H3]-16-Ketoestradiol

RL: SPN (Synthetic preparation); PREP (Preparation)

(stereoselective deuteration of estrogens using palladium/carbon as a catalyst)

RN 303128-12-3 CAPLUS

CN Estra-1,3,5(10)-trien-16-one-6,6,9-d3, 3,17-dihydroxy-, (17β) - (9CI) (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT:

THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 15 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

9

ACCESSION NUMBER: 2000

2000:549991 CAPLUS Full-text

DOCUMENT NUMBER: 134:147329

TITLE: Convenient synthesis of deuterated cycloalkanes from

polyhalophenols with nickel-aluminum alloy in alkaline

deuterium oxide

AUTHOR(S): Tsuzuki, Hirohisa; Mataka, Shuntaro; Tashiro, Masashi

CORPORATE SOURCE: Center of Advanced Instrumental Analysis, Kyusha

University, Kasuga, 816, Japan

SOURCE: Synthesis and Applications of Isotopically Labelled

Compounds 1997, Proceedings of the International

Symposium, 6th, Philadelphia, PA, United States, Sept.

14-18, 1997 (1998), Meeting Date 1997, 203-206.

Editor(s): Heys, J. Richard; Melillo, David G. John

Wiley & Sons Ltd.: Chichester, UK.

CODEN: 69AGFO

DOCUMENT TYPE: Conference

LANGUAGE: English

OTHER SOURCE(S): CASREACT 134:147329

AB A symposium report on the deuteration of polyhalophenolic substrates in the

presence of nickel-aluminum alloy.

CC 24-5 (Alicyclic Compounds)

Section cross-reference(s): 34

IT Deuteration

Deuteration catalysts

(deuterated cycloalkanes from polyhalophenols with nickel-aluminum

alloy in alkaline deuterium oxide)

IT 11114-68-4

RL: CAT (Catalyst use); USES (Uses)

(deuterated cycloalkanes from polyhalophenols with nickel-aluminum alloy in alkaline deuterium oxide)

IT 77787-72-5P 93131-17-0P, Cyclohexan-d11-ol

324520-34-5P 324520-35-6P 324520-36-7P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)

(deuterated cycloalkanes from polyhalophenols with nickel-aluminum alloy in alkaline deuterium oxide)

IT 11114-68-4

RL: CAT (Catalyst use); USES (Uses)

(deuterated cycloalkanes from polyhalophenols with nickel-aluminum alloy in alkaline deuterium oxide)

RN 11114-68-4 CAPLUS

CN Aluminum alloy, nonbase, Al, Ni (CA INDEX NAME)

Component Component

Registry Number

Al 7429-90-5

Ni 7440-02-0

IT 77787-72-5P 93131-17-0P, Cyclohexan-dll-ol

324520-34-5P 324520-35-6P 324520-36-7P

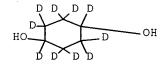
RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)

(deuterated cycloalkanes from polyhalophenols with nickel-aluminum alloy in alkaline deuterium oxide)

RN 77787-72-5 CAPLUS

CN 1,4-Cyclohexane-1,2,2,3,3,4,5,5,6,6-d10-diol (9CI) (CA INDEX NAME)



RN 93131-17-0 CAPLUS

CN Cyclohexan-dll-ol (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} D & D & D \\ \hline D & D & D \\ \hline \end{array} \begin{array}{c} D & D \\ \hline \end{array} \begin{array}{c} OH \\ \hline \end{array}$$

RN 324520-34-5 CAPLUS

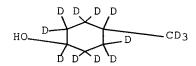
CN Cyclohexan-1,2,2,3,3,4,4,5,5,6-d10-o1, 6-(methyl-d3)- (9CI) (CA INDEX NAME)

RN 324520-35-6 CAPLUS

CN Cyclohexan-1,2,2,3,3,4,4,5,6,6-d10-ol, 5-(methyl-d3)- (9CI) (CA INDEX NAME)

RN 324520-36-7 CAPLUS

CN Cyclohexan-1,2,2,3,3,4,5,5,6,6-d10-ol, 4-(methyl-d3)- (9CI) (CA INDEX NAME)



REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 16 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2000:414551 CAPLUS Full-text

DOCUMENT NUMBER: 133:192893

TITLE: Multiple deuteration of water-soluble olefinic acids

with a [Pd(alizarin monosulfonate)2] catalyst

with a [rd(alizalin monosulionate/2] catalyst

AUTHOR(S): Papp, Eva; Banyai, Istvan; Joo, Ferenc

CORPORATE SOURCE: Research Group of Homogeneous Catalysis, Hungarian

Academy of Sciences, Debrecen, H-4010, Hung.

SOURCE: Reaction Kinetics and Catalysis Letters (2000), 69(1),

23-30

CODEN: RKCLAU; ISSN: 0304-4122

PUBLISHER: Akademiai Kiado

DOCUMENT TYPE: Journal LANGUAGE: English

AB Hydrogenations in aqueous systems with the soluble [Pd(alizarin monosulfonate)2] catalyst resulted in extensive deuteration of crotonic, trans-2-pentenoic and itaconic acids regardless of whether the deuterium source was D2 or D2O. Itaconic acid was deuterated up to 3.6 D/methylsuccinic

acid. Detailed 1H-and 13C-NMR studies identified six isotopomers of the deuterated methylsuccinic acid product and revealed an important role of the $\rm H/D$ exchange on the catalytically active Pd-intermediate.

CC 23-16 (Aliphatic Compounds)

IT Deuteration

Deuteration catalysts

Hydrogenation

Hydrogenation catalysts

(hydrogenation and multiple deuteration of water-soluble olefinic acids with a [Pd(alizarin monosulfonate)2] catalyst)

IT 74091-55-7

RL: CAT (Catalyst use); USES (Uses)

(hydrogenation and multiple deuteration of water-soluble olefinic acids with a [Pd(alizarin monosulfonate)2] catalyst)

IT 498-21-5P 169127-02-0P 289679-94-3P, Butanoic-d acid

289679-95-4P, Pentanoic-d acid 289679-96-5P 289679-97-6P 289679-98-7P 289679-99-8P

RL: SPN (Synthetic preparation); PREP (Preparation)

(hydrogenation and multiple deuteration of water-soluble olefinic acids with a [Pd(alizarin monosulfonate)2] catalyst)

IT 74091-55-7

RL: CAT (Catalyst use); USES (Uses)

(hydrogenation and multiple deuteration of water-soluble olefinic acids with a [Pd(alizarin monosulfonate)2] catalyst)

RN 74091-55-7 CAPLUS

CN Palladate(2-), bis[9,10-dihydro-3-hydroxy-4-(hydroxy-κ0)-9-oxo-10-(oxo-κ0)-2-anthracenesulfonato(2-)]-, disodium (9CI) (CA INDEX NAME)

IT 289679-94-3P, Butanoic-d acid 289679-95-4P, Pentanoic-d acid

RL: SPN (Synthetic preparation); PREP (Preparation)

(hydrogenation and multiple deuteration of water-soluble olefinic acids with a [Pd(alizarin monosulfonate)2] catalyst)

RN 289679-94-3 CAPLUS

CN Butanoic-d acid (9CI) (CA INDEX NAME)

. С— СН2— СН2— СН3

289679-95-4 CAPLUS RN

Pentanoic-d acid (9CI) (CA INDEX NAME) CN

HO_C_CH2_CH2_CH2_CH3

REFERENCE COUNT:

13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 17 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER:

1997:324308 CAPLUS Full-text

DOCUMENT NUMBER:

127:17291

TITLE:

First Evidence That the Mechanism of Catalytic

Hydrogenation with Homogeneous Palladium and Rhodium Catalysts Is Strongly Influenced by Substrate Polarity

AUTHOR(S):

Yu, Jinquan; Spencer, Jonathan B.

CORPORATE SOURCE:

University Chemical Laboratory, University of

Cambridge, Cambridge, CB2 1EW, UK

SOURCE:

Journal of the American Chemical Society (1997),

119(22), 5257-5258

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER:

American Chemical Society

DOCUMENT TYPE:

Journal

LANGUAGE: English

AB We have observed that when cis-alkenes are hydrogenated with homogeneous palladium and rhodium catalysts they readily isomerize to the transconfiguration with the incorporation of a deuterium atom. By studying how electron withdrawing and donating groups conjugated to the double bond influence the location of deuterium addition we have been able to gain a clear insight into how the metal hydrogen bond in the catalyst is polarized just prior to adding to the cis-alkene. Remarkably, the result demonstrate that the palladium hydrogen bond is capable of being polarized in either mode (a $Pd\delta+-H\delta-$ or b $Pd\delta--H\delta+$) depending on the coulombic properties of the substrate, whereas the rhodium catalyst studied is dominated by mode a $(Rh\delta+-H\delta-)$. This provides strong evidence that the mechanism of catalytic hydrogenation is a 2 electron process that can be dramatically affected by the substrate's

22-7 (Physical Organic Chemistry)

Section cross-reference(s): 29, 67

IT Conjugation (bond)

Deuteration

Deuteration catalysts

Electron transfer Hydrogenation Hydrogenation catalysts Isomerization Isomerization catalysts NMR (nuclear magnetic resonance) Regiochemistry

Resonance

Substituent effects

(strong substrate polarity effect on mechanism of catalytic hydrogenation with homogeneous palladium and rhodium catalysts)

IT 14694-95-2, Chlorotris(triphenylphosphine)rhodium
31277-98-2, Bis[1,2-bis(diphenylphosphino)ethane]palladium
RL: CAT (Catalyst use); PEP (Physical, engineering or chemical
process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent); USES

(strong substrate polarity effect on mechanism of catalytic hydrogenation with homogeneous palladium and rhodium catalysts)

IT 107-93-7P 140-10-3P, preparation 623-43-8P 943-89-5P 1005-64-7P 1011-54-7P 4747-15-3P **69104-43-4P** 89039-12-3P 89039-13-4P

RL: SPN (Synthetic preparation); PREP (Preparation)

(strong substrate polarity effect on mechanism of catalytic hydrogenation with homogeneous palladium and rhodium catalysts)

IT 14694-95-2, Chlorotris(triphenylphosphine)rhodium
31277-98-2, Bis[1,2-bis(diphenylphosphino)ethane]palladium
RL: CAT (Catalyst use); PEP (Physical, engineering or chemical
process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent); USES
(Uses)

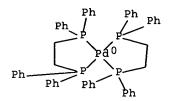
(strong substrate polarity effect on mechanism of catalytic hydrogenation with homogeneous palladium and rhodium catalysts)

RN 14694-95-2 CAPLUS

CN Rhodium, chlorotris(triphenylphosphine)-, (SP-4-2)- (CA INDEX NAME)

RN 31277-98-2 CAPLUS

CN Palladium, bis[1,2-ethanediylbis[diphenylphosphine-κΡ]]-, (T-4)(9CI) (CA INDEX NAME)



IT 69104-43-4P

RL: SPN (Synthetic preparation); PREP (Preparation)

(strong substrate polarity effect on mechanism of catalytic hydrogenation with homogeneous palladium and rhodium catalysts)

RN 69104-43-4 CAPLUS

CN 2-Propenoic-2-d acid, 3-phenyl- (9CI) (CA INDEX NAME)

D HO2C-C-CH-Ph

REFERENCE COUNT: 20 THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 18 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1996:98985 CAPLUS Full-text

DOCUMENT NUMBER: 124:260517

TITLE: Retention of optical purity in H-D exchange reactions

catalyzed by cobalt-aluminum alloy in Na2CO3-D2O

AUTHOR(S): Mukumoto, Mamoru; Tsuzuki, Hirohisa; Mataka, Shuntaro;

Tashiro, Masashi; Tsukinoki, Takehito; Nagano,

Yoshiaki

CORPORATE SOURCE: Dep. Mol. Sci. Technol., Kyushu Univ., Kasuga, 816,

Japan

SOURCE: Chemistry Letters (1996), (2), 165-6

CODEN: CMLTAG; ISSN: 0366-7022

PUBLISHER: Nippon Kagakkai

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 124:260517

AB Co-Al alloy in a sodium carbonate-deuterium oxide solution catalyzes the H-D exchange reaction of optically active benzylic hydrogen atom without racemization. Thus, (R)-mandelic acid give Me α -D-(R)-mandelate in 89% yield with 99% enantiomeric excess.

CC 25-18 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)

IT Deuteration catalysts

(stereoselective; deuteration of benzylic compds. with retention of configuration using Co-Al alloy in Na2CO3-D2O)

IT Deuteration

(stereoselective, deuteration of benzylic compds. with retention of configuration using Co-Al alloy in Na2CO3-D2O)

IT 11114-55-9

RL: CAT (Catalyst use); USES (Uses)

(deuteration of benzylic compds. with retention of configuration using Co-Al alloy in Na2CO3-D2O)

IT 175289-31-3P **175289-32-4P** 175289-33-5P 175289-34-6P 175289-35-7P

RL: SPN (Synthetic preparation); PREP (Preparation)

(deuteration of benzylic compds. with retention of configuration using Co-Al alloy in Na2CO3-D2O)

IT 11114-55-9

RL: CAT (Catalyst use); USES (Uses)

(deuteration of benzylic compds. with retention of configuration using Co-Al alloy in Na2CO3-D2O)

RN 11114-55-9 CAPLUS

CN Aluminum alloy, nonbase, Al,Co (CA INDEX NAME)

Component Component
Registry Number

Al 7429-90-5 Co 7440-48-4

IT 175289-32-4P

RL: SPN (Synthetic preparation); PREP (Preparation)

(deuteration of benzylic compds. with retention of configuration using Co-Al alloy in Na2CO3-D2O)

RN 175289-32-4 CAPLUS

CN Benzeneacetic-d acid, α -methyl-, methyl ester, (R)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

L19 ANSWER 19 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 1989:231056 CAPLUS Full-text

DOCUMENT NUMBER: 110:231056

TITLE: Homogeneous deuteration of alkenes using

[RhCl(4R,5R-diop)] catalysts

AUTHOR(S): Gungor, Muammer; Jardine, Fred H.; Wheatley, J. Denis

CORPORATE SOURCE: Dep. Phys. Sci., North East London Polytech., London,

E15 4LZ, UK

SOURCE: Polyhedron (1988), 7(19-20), 1827-9

CODEN: PLYHDE; ISSN: 0277-5387

DOCUMENT TYPE: Journal LANGUAGE: English

Mass spectrometric analyses of the products from the homogeneous deuteration of alkenes using the title (4R,5R- DIOP)/{[RhCl(cyclooctene)2]2} catalyst system show that considerable quantities of polydeuterated products are obtained. These products arise from the decomposition of the intermediate rhodium(III)alkyl complex [RhDCl(alkyl)(DIOP)] before the second atom of deuterium can be transferred to the alkyl ligand. Its decomposition by β -hydride abstraction brings about both polydeuteration and scrambled addition of deuterium to the alkene. The yields of specifically deuterated products are inferior to those obtained from Wilkinson-type catalysts. The best yields of dideuterated products are obtained from substituted alkenes that chelate to the catalyst and thereby stabilize the intermediate alkyl. The preparation of a threitol ditosylate intermediate in dry pyridine was noted for its dangerous exothermicity.

CC 23-2 (Aliphatic Compounds)

Section cross-reference(s): 24, 25

IT Deuteration

(of alkenes, mechanism of catalytic)

IT Deuteration catalysts

(rhodium-DIOP complex, for alkenes)

IT 12279-09-3

RL: CAT (Catalyst use); USES (Uses)

(catalysts, containing DIOP, for deuteration of alkenes)

IT 66502-82-7P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(preparation and carbon-13 NMR of)

IT 292-64-8P, Cyclooctane 24588-43-0P, Octane-1,2-d2 33283-80-6P, Bicyclo[2.2.1]heptane-2,3-d2 71501-04-7P, Octane-d 73811-48-0P 86812-02-4P, Cyclooctane-d 92475-64-4P, Hexane-1,2-d2 95236-97-8P, Heptane-1,2-d2 95237-00-6P 98821-91-1P, Cyclohexane-1,2-d2

118296-70-1P, Octane-1,2,?-d3 118296-71-2P, Octane-1,2,?,?-d4 118297-06-6P 118297-07-7P 118297-11-3P, Heptane-d 118297-12-4P, Heptane-1,2,?-d3 118297-17-9P, Hexane-1,2,?-d3 118297-18-0P, Hexane-1, 2, ?, ?-d4 120625-93-6P, Cyclohexanone-2,3-d2 120625-94-7P, Cyclooctane-1,2-d2 120625-96-9P **120625-97-0P**, 120625-95-8P 120625-98-1P, Cyclohexane-1,2,?-d3 2-Hexanone-5,6-d2 120625-99-2P, 120626-00-8P, Cyclooctane-1,2,?-d3 Cyclohexanone-2,3,?-d3 120626-01-9P 120626-02-0P **120626-03-1P** 120626-04-2P, Cyclooctane-1,2,?,?-d4 120626-05-3P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

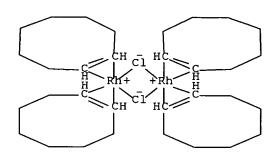
IT 12279-09-3

RL: CAT (Catalyst use); USES (Uses)

(catalysts, containing DIOP, for deuteration of alkenes)

RN 12279-09-3 CAPLUS

CN Rhodium, di- μ -chlorotetrakis[(1,2- η)-cyclooctene]di- (CA INDEX NAME)



IT 66502-82-7P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(preparation and carbon-13 NMR of)

RN 66502-82-7 CAPLUS

CN Butanoic-2,3-d2 acid, methyl ester (CA INDEX NAME)

IT 120625-97-0P, 2-Hexanone-5,6-d2 120626-03-1P 120626-05-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 120625-97-0 CAPLUS

CN 2-Hexanone-5,6-d2 (9CI) (CA INDEX NAME)

$$\begin{array}{c} D \\ DCH_2 - CH_2 - CH_2 - CH_2 - C \\ \end{array}$$

120626-03-1 CAPLUS RN

Butanoic-2,3,4-d3 acid, methyl ester (9CI) (CA INDEX NAME) CN

120626-05-3 CAPLUS RN

Butanoic-2,3,4,?-d4 acid, methyl ester (9CI) (CA INDEX NAME) CN

L19 ANSWER 20 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER:

1982:545159 CAPLUS Full-text

DOCUMENT NUMBER:

97:145159

ORIGINAL REFERENCE NO.: 97:24193a,24196a

TITLE:

Some stereochemical characteristics of C-1H-C-2H exchange-reactions with Raney nickel catalyst in

deuterium oxide

AUTHOR(S):

SOURCE:

Balza, Felipe; Perlin, Arthur S.

CORPORATE SOURCE:

Dep. Chem., McGill Univ., Montreal, QC, H3C 3G1, Can.

Carbohydrate Research (1982), 107(2), 270-8 CODEN: CRBRAT; ISSN: 0008-6215

Journal

DOCUMENT TYPE:

LANGUAGE: English

In C-deuteration of carbohydrates with Raney Ni in D2O, the following reaction-characteristics were observed: (a) at least 2 OH groups, not necessarily contiguous, are required for 1H-2H exchange to occur; (b) the rate of isotope incorporation into alkyl glycopyranosides is relatively slow when the C-H bond undergoing exchange is syn-axial with respect to an alkoxy group or when the bond is axial and vicinal to an equatorial alkoxyl group; (c) an increase in the size of the alkoxyl substituent leads to further diminution in the rate of exchange; (d) an equatorial H atom undergoes exchange more readily than an axial one; and (e) isotope exchange proceeds primarily with retention of configuration, although some degree of isomerization is almost always observed

33-1 (Carbohydrates) CC

Section cross-reference(s): 22

IT Deuteration catalysts

(nickel, for carbohydrates)

IT Deuteration

(of carbohydrates over Raney nickel)

7440-02-0, uses and miscellaneous

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for deuteration of carbohydrates)

IT 68922-38-3P 83158-38-7P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

IT 7440-02-0, uses and miscellaneous

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for deuteration of carbohydrates)

RN 7440-02-0 CAPLUS

CN Nickel (CA INDEX NAME)

Νi

IT 83158-38-7P

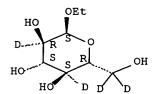
RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 83158-38-7 CAPLUS

CN α -D-Glucopyranoside-2,4,6,6-C-d4, ethyl (9CI) (CA INDEX NAME)

Absolute stereochemistry.



L19 ANSWER 21 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1981:461411 CAPLUS Full-text

DOCUMENT NUMBER: 95:61411

ORIGINAL REFERENCE NO.: 95:10371a,10374a

TITLE: A versatile procedure for the preparation of palmitic

acid-d2 and stearic acid-d6 Adlof, R. O.; Emken, E. A.

CORPORATE SOURCE: North. Reg. Res. Cent., USDA, Peoria, IL, 61604, USA
SOURCE: Journal of Labelled Compounds and Radiopharmaceuticals

(1981), 18(3), 419-26

CODEN: JLCRD4; ISSN: 0362-4803

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 95:61411

Me(CH2)3(CD2)2(CHD)2(CH2)7CO2H and Me(CH2)5(CHD)2(CH2)7CO2H were prepared from the corresponding 9,10-unsatd. Me esters by catalytic deuteration [(Ph3P)3RhCl, C6H6, under D, overnight) and saponification (KOH, aqueous MeOH, under N, reflux, 1.5). Me(CH2)5CH:CH(CH2)7CO2Me was prepared (65%) by Wittig reaction of [Me(CH2)6Ph3P]+I- with OHC(CH2)7CO2Me (DMF, NaOMe, 11°, overnight).

CC 23-16 (Aliphatic Compounds)

IT Deuteration

AUTHOR(S):

(of unsatd. fatty acid esters, rhodium-catalyzed)

IT Deuteration catalysts

(tris(triphenylphosphine)chlororhodium, for unsatd. fatty acids)

IT 14694-95-2

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for deuteration of unsatd. fatty acids)

IT 78387-67-4P 78387-69-6P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)
 (preparation and saponification of)

IT 78387-68-5P 78387-70-9P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

IT 14694-95-2

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for deuteration of unsatd. fatty acids)

RN 14694-95-2 CAPLUS

CN Rhodium, chlorotris(triphenylphosphine)-, (SP-4-2)- (CA INDEX NAME)

IT 78387-67-4P 78387-69-6P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)

(preparation and saponification of)

RN 78387-67-4 CAPLUS

CN Octadecanoic-9,10,13,13,14,14-d6 acid, methyl ester (9CI) (CA INDEX NAME)

RN 78387-69-6 CAPLUS

CN Hexadecanoic-9,10-d2 acid, methyl ester (9CI) (CA INDEX NAME)

IT 78387-68-5P 78387-70-9P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 78387-68-5 CAPLUS

CN Octadecanoic-9,10,13,13,14,14-d6 acid (9CI) (CA INDEX NAME)

RN 78387-70-9 CAPLUS

CN Hexadecanoic-9,10-d2 acid (9CI) (CA INDEX NAME)

D D | He-(CH₂)₅-CH-CH-(CH₂)₇-CO₂H

L19 ANSWER 22 OF 22 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1977:139371 CAPLUS Full-text

DOCUMENT NUMBER: 86:139371

ORIGINAL REFERENCE NO.: 86:21873a,21876a

TITLE: Molecular hydrogenation, deuteration and oxygenation

AUTHOR(S): Khan, N. A.; Ahmed, R.

CORPORATE SOURCE: BCSIR Lab., Chittagong, Bangladesh

SOURCE: Bangladesh Journal of Scientific and Industrial

Research (1976), 11(1-4), 148-53 CODEN: BJSIBL; ISSN: 0304-9809

DOCUMENT TYPE: Journal LANGUAGE: English

AB Four acetylenic compds. and Me linoleate were hydrogenated and deuterated using H and D containing Ni W8, W9, and W10 catalysts, e.g., Me(CH2)7C.tplbond.C(CH2)7CO2Me gave Me(CH2)7CH:CH(CH2)7CO2Me and Me(CH2)7CD:CD(CH2)7CO2Me. Oxygenation of the reduced products showed that the deutero-compds. have higher induction period than the corresponding H compds.

CC 23-17 (Aliphatic Compounds)

IT Deuteration catalysts

Hydrogenation catalysts

(nickel, for acetylenic compds. and methyl linolenate)

IT Deuteration

Hydrogenation

(of acetylenic compds. and methyl linolenate)

IT 7440-02-0, uses and miscellaneous

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for hydrogenation and deuteration of acetylenic compds. and methyl linolenate)

IT 2462-84-2P 2462-85-3P 5557-31-3P 56554-40-6P

62439-46-7P 62439-47-8P 62439-48-9P 62439-49-0P

62439-50-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation and oxygenation of)

IT 7440-02-0, uses and miscellaneous

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for hydrogenation and deuteration of acetylenic compds. and methyl linolenate)

RN 7440-02-0 CAPLUS

CN Nickel (CA INDEX NAME)

Νi

IT 56554-40-6P 62439-46-7P 62439-50-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation and oxygenation of)

RN 56554-40-6 CAPLUS

CN 9-Octadecenoic-9,10-d2 acid, methyl ester (9CI) (CA INDEX NAME)

$$MeO - \stackrel{O}{U} - (CH_2)_7 - \stackrel{D}{C} = \stackrel{D}{C} - (CH_2)_7 - Me$$

RN 62439-46-7 CAPLUS

CN 9,12-Octadecadienoic-9,10,12,13-d4 acid, methyl ester (9CI) (CA INDEX NAME)

$$Me - (CH_2)_4 - C = C - CH_2 - C = C - (CH_2)_7 - C - OMe$$

RN 62439-50-3 CAPLUS

CN 9,12-Octadecadienoic-15,16-d2 acid, methyl ester (9CI) (CA INDEX NAME)

Compound of Claim 12:

=> d que 132

L25 3 SEA FILE=REGISTRY ABB=ON PLU=ON 13380-89-7/CRN

L26 2 SEA FILE=CAPLUS ABB=ON PLU=ON L25

L27 STR

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 11

STEREO ATTRIBUTES: NONE

L28 52 SEA FILE=REGISTRY FAM FUL L27

L29 133 SEA FILE=CAPLUS ABB=ON PLU=ON L28

L30 168165 SEA FILE=CAPLUS ABB=ON PLU=ON ?DEUTER?
L31 4 SEA FILE=CAPLUS ABB=ON PLU=ON L29 AND L30
L32 6 SEA FILE=CAPLUS ABB=ON PLU=ON L31 OR L26

=> d 132 ibib abs hitind hitstr tot

L32 ANSWER 1 OF 6 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:589514 CAPLUS Full-text

DOCUMENT NUMBER: 141:139883

TITLE: Method of catalytic deuteration of carbonyl

compounds or secondary alcohols by heavy water

INVENTOR(S): Ito, Nobuhiro; Maesawa, Tsuneaki; Muto, Kazushige;

Hirota, Kosaku; Sajiki, Hironao

PATENT ASSIGNEE(S): Wako Pure Chemical Industries, Ltd., Japan

SOURCE: PCT Int. Appl., 42 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.					KIN	D	DATE			APPL	ICAT	ION I	NO.		D	ATE		
							_									_		
	WO	2004	0608	31		A1		2004	0722	1	WO 2	003-	JP14	182		2	0031	107
		W:	ΑE,	AG,	AL,	AM,	AT,	ΑU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	BZ,	CA,	CH,	CN,
			CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,
			GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	KP,	KR,	KZ,	LC,	LK,	LR,
			LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NI,	NO,	NZ,	OM,
			PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,	TJ,	TM,	TN,
			TR,	TT,	TZ,	UA,	ŪG,	US,	UΖ,	VC,	VN,	YU,	ZA,	ZM,	ZW			
		RW:	GH,	GM,	KE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZM.	ZW.	AM.	AZ.	BY.

```
KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES,
             FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR,
             BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
     CA 2511885
                          A1
                                20040722
                                            CA 2003-2511885
                                                                    20031107
     AU 2003277596
                          A1
                                20040729
                                            AU 2003-277596
                                                                    20031107
     EP 1577280
                          Α1
                                20050921
                                            EP 2003-814536
                                                                    20031107
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
     CN 1732135
                          Α
                                20060208
                                            CN 2003-80107483
                                                                    20031107
     US 2006116535
                          A1
                                20060601
                                            US 2005-539188
                                                                    20050616
     IN 2005KN01449
                                20070720
                                            IN 2005-KN1449
                          Α
                                                                    20050726
PRIORITY APPLN. INFO.:
                                            JP 2002-378932
                                                                A 20021227
                                            WO 2003-JP14182
                                                                   20031107
                                                                W
OTHER SOURCE(S):
                         CASREACT 141:139883; MARPAT 141:139883
     Described is a method of deuterating a carbonyl or secondary alc. compound
AΒ
     represented by the general formula R1-X-R2 (I) (wherein R1 = alkyl optionally
     possessing a CH:CH or C.tplbond.C bond, aralkyl; R2 = alkyl optionally
     possessing a CH:CH or C.tplbond.C bond, aryl, aralkyl, alkoxy, aryloxy,
     hydroxy; X carbonyl, hydroxymethylene), which comprises reacting the compound
     represented by the general formula I with a deuterium source, in particular
     D20, in the presence of a catalyst selected among activated palladium,
     platinum, rhodium, ruthenium, nickel, and cobalt catalysts. By the method,
     deuteration, which has been conducted under severe conditions, can be
     conducted under neutral conditions. Even when the compound contains an
     unsatd. bond, it can be deuterated without reducing the unsatd. bond. Not
     only hydrogens near the carbonyl or hydroxymethylene group but also those
     remotely situated from these groups are selectively deuterated without
     deuterating the carbon-carbon double or triple bonds. Thus, 500 mg
     tricyclo[5.2.1.02'6]decan-8-ol and 100 mg Pd-C were suspended in 17 mL D20,
     purged with H, and heated at 180° for 24 h in an oil bath to give
     tricyclo[5.2.1.02'6]decan-8-ol deuterated by 96% at 8-position and 88% at
     other positions.
IC
     ICM C07B059-00
     ICS
         C07C029-00; C07C031-02; C07C035-08; C07C035-29; C07C035-37;
         C07C045-00; C07C049-04; C07C049-08; C07C049-433; C07C049-453;
         C07C051-00; C07C053-10; C07C053-124; C07C057-04; C07M005-00
CC
    21-2 (General Organic Chemistry)
ST
    deuterated carbonyl compd secondary alc prepn; catalytic deuteration
    ketone carboxylic acid secondary alc; palladium platinum deuteration
    catalyst; heavy water deuteration carbonyl compd secondary alc
ΙT
    Deuteration
      Deuteration catalysts
        (catalytic deuteration of carbonyl compds. or secondary alc.
       compds. with heavy water in presence of palladium, platinum, rhodium,
        ruthenium, or nickel)
IT
    Carbonyl compounds (organic), reactions
    Carboxylic acids, reactions
    Ketones, reactions
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (catalytic deuteration of carbonyl compds. or secondary alc.
       compds. with heavy water in presence of palladium, platinum, rhodium,
        ruthenium, or nickel)
IT
    Alcohols, reactions
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (secondary; catalytic deuteration of carbonyl compds. or
       secondary alc. compds. with heavy water in presence of palladium,
       platinum, rhodium, ruthenium, or nickel)
IT
    7440-02-0, Raney nickel, uses
    RL: CAT (Catalyst use); USES (Uses)
        (catalysts; catalytic deuteration of carbonyl compds. or
```

secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) ΙT 7440-05-3, Palladium, uses 7440-05-3D, Palladium, supported on carbon 7440-06-4, Platinum, uses 7440-06-4D, Platinum, supported on carbon 7440-16-6, Rhodium, uses 7440-16-6D, Rhodium, supported on alumina 7440-16-6D, Rhodium, supported on carbon 7440-18-8, Ruthenium, uses .7440-18-8D, Ruthenium, supported on carbon 7440-48-4, Cobalt, uses RL: CAT (Catalyst use); USES (Uses) (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) IT 67-64-1, Acetone, reactions 78-93-3, 2-Butanone, reactions 79-31-2, Isobutyric acid 79-41-4, Methacrylic acid, reactions 106-35-4, 3-Heptanone 108-93-0, Cyclohexanol, reactions 108-94-1, Cyclohexanone, reactions 110-43-0, 2-Heptanone 123-19-3, 4-Heptanone 127-09-3, Sodium acetate 497-38-1, 2-Norbornanone 543-49-7, 2-Heptanol 589-55-9, 4-Heptanol 3385-61-3, Tricyclo[5.2.1.02,6]-3-decen-8-ol 5536-61-8, Sodium methacrylate 7789-20-0, Water-d2 **13380-89-7**, Tricyclo[5.2.1.02,6]decan-8-ol 13380-94-4, Tricyclo[5.2.1.02,6]decan-8-63870-91-7, Norbornenol RL: RCT (Reactant); RACT (Reactant or reagent) (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) 79-31-2DP, Isobutyric acid, deuterated IT 108-93-0DP. Cyclohexanol, deuterated 666-52-4P, 2-Propanone-1,1,1,3,3,3-d6 13380-89-7DP, Tricyclo[5.2.1.02,6]decan-8-ol, deuterated 14044-94-1P 18153-61-2DP, Bicyclo[2.2.1]heptan-2-one-3,3-d2, deuterated 21273-02-9DP, Cyclohexan-1-d-ol, deuterated 51209-49-5P, Cyclohexanone-d10 53481-06-4P 55935-44-9P 63870-91-7DP, Norbornenol, deuterated 64118-21**-**4P 91468-78-9DP, Bicyclo[2.2.1]heptan-2-d-2-ol, deuterated 350820-09-6P 725242-18-2P, 4-Heptanone-d14 725242-19-3DP, **deuterated** 725242-21-7DP, **deuterated** 725242-22-8P, 2-Heptanone-d14 725242-23-9P, 3-Heptanone-d14 725242-24-0P, 2-Heptanol-d15 725242-25-1P, 4-Heptanol-d15 725242-26-2DP, **deuterated** 725242-27-3DP, **deuterated** 725242-28-4DP, **deuterated** 725242-29-5DP, deuterated 725242-29-5P 725242-30-8P 725242-31-9P 725242-32-0DP, **deuterated** RL: SPN (Synthetic preparation); PREP (Preparation) (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium, ruthenium, or nickel) 13380-89-7, Tricyclo[5.2.1.02,6]decan-8-ol IT RL: RCT (Reactant); RACT (Reactant or reagent) (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, rhodium,

RN

CN

IT

ruthenium, or nickel)

13380-89-7 CAPLUS

4,7-Methano-1H-inden-5-ol, octahydro- (CA INDEX NAME)

725242-29-5DP, deuterated 725242-29-5P 725242-30-8P

RL: SPN (Synthetic preparation); PREP (Preparation) (catalytic deuteration of carbonyl compds. or secondary alc. compds. with heavy water in presence of palladium, platinum, ruthenium, or nickel)

RN 13380-89-7 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro- (CA INDEX NAME)

HO

RN 725242-29-5 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-5-d- (9CI) (CA INDEX NAME)

DHO

RN 725242-29-5 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-5-d- (9CI) (CA INDEX NAME)

DHO D

RN 725242-30-8 CAPLUS

CN 4,7-Methano-1H-inden-1,1,2,3,6,7,7-d7-5-ol, octahydro-2,3,3a,4,5,6,7,7a-d8-(9CI) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L32 ANSWER 2 OF 6 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1990:199410 CAPLUS Full-text

DOCUMENT NUMBER:

112:199410

TITLE:

 β -Methyl- δ -valerolactone adducts

INVENTOR(S):

Kuroki, Masayuki; Yokoshima, Minoru; Maeda, Toshihiko;

Yoshimura, Noriaki

PATENT ASSIGNEE(S):

Nippon Kayaku Co., Ltd., Japan; Kuraray Co., Ltd.

SOURCE:

Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01258645	Α	19891016	JP 1988-83125	19880406
PRIORITY APPLN. INFO.:			JP 1988-83125	19880406
GI				

- AB R(COCH2CHMeCH2CH2O)nH [R = R1C6H4O(CH2CHR2O)m, CpH2p+10, perhydronaphthyloxy, perhydromethanoindanyloxy; R1 = H, C1-12 alkyl; R2 = H, Me; average m = 1-10; average n = 0.5-10; p = 1-20], whose (meth)acrylate esters are useful as diluents in UV-curable coatings, are prepared Thus, 374 parts poly(oxyethylene) p-nonylphenyl ether was treated with 684 parts β -methyl- δ -valerolactone at room temperature for 20 h to give 830 parts product with OH value 67.6.
- IC ICM C07C069-675
- CC 35-7 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 42
- IT 126161-91-9P 126161-92-0P **126161-93-1P** 126161-94-2P 126286-45-1P 126286-48-4P 126305-36-0P
 - RL: PREP (Preparation)

- IT 126161-93-1P
 - RL: PREP (Preparation)

(preparation of, as intermediate for methacrylate diluents, for UV-curable coatings)

- RN 126161-93-1 CAPLUS
- CN 2H-Pyran-2-one, tetrahydro-4-methyl-, homopolymer, octahydro-4,7-methano-1H-inden-5-yl ester (9CI) (CA INDEX NAME)

CM 1

CRN 13380-89-7 CMF C10 H16 O

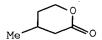
CM 2

CRN 97145-14-7

CMF (C6 H10 O2)x CCI PMS

> CM 3

CRN 1121-84-2 CMF C6 H10 O2



L32 ANSWER 3 OF 6 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1974:81650 CAPLUS Full-text

DOCUMENT NUMBER: 80:81650

ORIGINAL REFERENCE NO.: 80:13137a,13140a

TITLE: Dehydration of 5-hydroxytetrahydro-exo-

dicyclopentadiene with acid

Gates, Marshall; Zabriskie, John L., Jr. AUTHOR(S):

CORPORATE SOURCE: Dep. Chem., Univ. Rochester, Rochester, NY, USA SOURCE: Journal of Organic Chemistry (1974), 39(2), p 222-7

CODEN: JOCEAH; ISSN: 0022-3263

DOCUMENT TYPE: Journal LANGUAGE: English

GΙ For diagram(s), see printed CA Issue.

AΒ A study of the dehydration of 3a-deuterio-5-hydroxytetrahydro-exodicyclopentadiene (I) with acid suggests that the formation of 5.6-dihydroexo-dicyclopentadiene (II) proceeds through 2,3-dihydro-exo-dicyclopentadiene (III) or the equilibrating ions formed from the latter by protonation followed by a 1,3-hydride shift, a 1,2-hydride shift, and proton loss.

CC 22-3 (Physical Organic Chemistry)

IT 42913-50-8

> RL: RCT (Reactant); RACT (Reactant or reagent) (dehydration of)

IT 10271-44-0

RL: RCT (Reactant); RACT (Reactant or reagent)

(dehydration of, mechanism of)

IT 42913-50-8

RL: RCT (Reactant); RACT (Reactant or reagent) (dehydration of)

RN 42913-50-8 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-3a-d-, $(3a\alpha, 4\beta, 5\beta, 7.b)$ eta., $7a\alpha$) - (9CI) (CA INDEX NAME)

Relative stereochemistry.

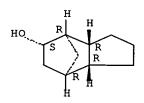
IT 10271-44-0

RL: RCT (Reactant); RACT (Reactant or reagent)
 (dehydration of, mechanism of)

RN 10271-44-0 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-, (3aR,4R,5S,7R,7aR)-rel- (CA INDEX NAME)

Relative stereochemistry.



L32 ANSWER 4 OF 6 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1971:32153 CAPLUS Full-text

DOCUMENT NUMBER: 74:32153
ORIGINAL REFERENCE NO.: 74:5169a,5172a

TITLE: Polymerization of α -olefins

INVENTOR(S): Schmitt, Karl; Gude, Fritz; Samblebe, Reinhard

PATENT ASSIGNEE(S): Veba-Chemie A.-G. SOURCE: Ger. Offen., 22 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 1918995	Α	19701112	DE 1969-1918995	19690415
AT 301169	В	19720825	AT 1970-1008	19700204
GB 1304093	Α	19730124	GB 1970-7318	19700216
BE 746667	Α	19700731	BE 1970-746667	19700227
NL 7003130	Α	19701019	NL 1970-3130	19700305
FR 2039082	A5	19710108	FR 1970-10329	19700323
PRIORITY APPLN. INFO.:			DE 1969-1918995 A	19690415

AB Ti(OBu)4, Ti(OPr)4, titanyl acetylacetonate, or a similar Ti compound, is used with VO(OBu)3, V acetylacetonate, or a similar V compound and with Et2AlCl, EtAlCl2, Et2AlBr, iso-Pr2AlBr, or a similar Al compound, optionally containing Et3Al and ZnCl2, Et2Zn, Et2Cd, or a similar compound, to prepare polymerization catalysts for C2H4, propylene, 4-methyl-1-pentene, and similar α -olefins. The catalysts have high catalytic activity, give colorless polymers even if the catalyst residue is not washed from the polymer, and give powdered polyolefins having good flow properties. Thus, 1 mole of a 1:3 molar mixture of Ti(OBu)4 and VO(OBu)3 in light petroleum was treated with 5.5 moles Et2AlCl to prepare a catalyst.

- IC C08F001-42
- CC 35 (Synthetic High Polymers)
- IT 17501-79-0 19059-01-9 24742-16-3 28860-26-6 30860-71-0 32673-47-5 32673-53-3, Malonic acid, titanium(4+) salt (2:1)

32673-54-4 32719-33-8

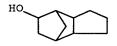
RL: CAT (Catalyst use); USES (Uses) (catalysts, for polyn. of olefins)

IT 32719-33-8

RL: CAT (Catalyst use); USES (Uses) (catalysts, for polyn. of olefins)

RN 32719-33-8 CAPLUS

CN 4,7-Methanoindan-5-ol, hexahydro-, titanium(4+) salt, exo- (8CI) (CA INDEX NAME)



●1/4 Ti(IV)

L32 ANSWER 5 OF 6 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER:

1964:432551 CAPLUS Full-text

DOCUMENT NUMBER:

61:32551

ORIGINAL REFERENCE NO.:

61:5679e-h,5680a-b

TITLE:

Oxymercuration of endo- and exo-dicyclopentadienes

AUTHOR(S):

Stille, J. K.; Stinson, S. C.

CORPORATE SOURCE:

Univ. of Iowa, Iowa City

SOURCE:

Tetrahedron (1964), 20(6), 1387-95

CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE:

Journal

LANGUAGE:

AB

Unavailable

OTHER SOURCE(S):

CASREACT 61:32551

GI For diagram(s), see printed CA Issue.

cf. Traylor and Baker, CA 59, 10101g. To 2.2 g. LiAlH4 in 15 ml. Et20 was added slowly 10.35 g. octahydro-exo-4,7-methanoinden-5-one in 20 ml. Et20 and the whole refluxed 1 hr. to give 6 g. octahydro-exo-4,7methanoinden- endo-5-ol (I), b1.5 73°, n19.5D 1.5127. The procedure of Cristol (C., et al., CA 54, 20908e) with I gave 52% octahydro-endo-methoxy-exo-4,7- methanoindene, b9 81°, n19.5D 1.4860. Hydroboration of 2,3,3a,4,7,7a-hexahydro-endo-4,7methanoindene gave octahydro-endo-4,7- methanoinden-exo-5-ol (II), m. 79.5-81° (MeNO2). II, as above, gave 47% octahydroexo-5-methoxy-endo-4,7-methanoindene (IIa), b6 84-6°, n19.5D 1.4923; CrO3 oxidation of II in C5H5N gave the ketone, which was reduced with LiAlH4 to give octahydro-endo-4,7-methanoinden- 5-ol (III), m. 119.5-20° (petr. ether). III gave 52% octahydroendo-5-methoxy-4,7methanoindene, b9 81°, n19.5D 1.4860. HgCl2 and endo-dicyclopentadiene (IV) in MeOH gave 79% of the methoxymercuric chloride adduct (V), m. 133°; the same reaction with exo-cyclopentadiene (Va) proceeded twice as rapidly to give 50% of the adduct (Vb), m. $135-7^{\circ}$. Hg(OAc)2 (35 g.), 100 ml. H2O, and 10.7 g. IVshaken 10 hrs. gave 42% hydroxymercuric acetate adduct (VI), m. 129-31° (CCl4); a similar reaction with V was faster and gave 42% unstable adduct (VIa), m. $130-5^{\circ}$ (CCl4). To 40 g. V in 400 ml. H2O, with agitation, was added in small portions 152 g. 3% Na-Hg, the whole extracted with Et2O, and the Et2O extract dried and concentrated gave, via vapor phase chromatography a 1:1 mixture (VII) of IV and VIII; pure VIII (9.7 g.) was obtained by distillation, b7 77-8°, n20D 1.5007. Similarly, Vb gave 50% IX, b9 81-3°, n19.5D 1.4976. VI gave 33% X, b4 91°, m. 42-4° (MeNO2) (p-nitrobenzoate m. 121.5-2.5°), and VIa

gave 12.5% XI, b3-5 92°, n19.5D 1.5262 (p-nitrobenzoate m. 126.5-8.5°). VIII (5.65 g.), EtOH, and PtO2 gave 3.6 g. IIa. Similarly, 4.8 g. IX gave 3.2 g. octahydroexo-5-methoxy-exo-4,7- methanoindene (XII), b7 81, n20D 1.4855, 2.28 g. X gave 1. 46 g. II, and 4.8 g. XI gave 1.58 g. octahydro-exo-4,7- methanoinden-exo-5-ol (XIII), m. 50.5-51°. V (20.4 g.), 200 ml. D2O, and 114 g. 3% Na-Hg gave 1.27 g. octahydro-exo-5-deutero-exo-6-methoxy-endo-4,7- methanoindene, b6 85-6°. The nuclear magnetic resonance spectra of some of these derivs. are discussed, the conclusion being that the mercuri group is cis or exo to the MeO or He groups.

CC 39 (Organometallic and Organometalloidal Compounds)

77-80-5P, Tin, thiobis[triphenyl-IT 133-21-1P, 4,7-Methanoinden-5-ol, 3a, 4, 5, 6, 7, 7a-hexahydro-(?), stereoisomers 338-48-7P, Germane, tributylfluoro 358-41-8P, Germane, triethylfluoro 994-28-5P, Germane, chlorotriethyl-1067-10-3P, Germane, bromotriethyl-1112-65-8P, Germane, acetoxytributyl 1441-22-1P, Tin, triphenyl(phenylthio)-2117-36-4P, Germane, tributylchloro 2290-67-7P, Germane, triethylisothiocyanato-2587-86-2P, Germane, oxybis[tributyl 10271-44-0P, 4,7-Methanoindan-5-ol, hexahydro-, exo-exo-13314-51-7P, Germane, triethyliodo- 22998-59-0P, Germane, dichlorodiisopropyl 24692-20-4P, Germanecarbonitrile, triethyl 53018-24-9P, 4,7-Methanoindene, 3a,4,5,6,7,7a-hexahydro-5-methoxy-, stereoisomers 57831-55-7P, Germane, triethylisocyanato-57879-96-6P, Tin, (methylthio)triphenyl- 79851-34-6P, Tin, (2-naphthylthio)triphenyl-92168-27-9P, 4,7-Methanoinden-5-ol, 6-(acetoxymercuri)-3a,4,5,6,7,7ahexahydro-, stereoisomers 93818-38-3P, 4,7-Methanoinden-5-ol, 3a,4,5,6,7,7a-hexahydro-(?), p-nitrobenzoate, stereoisomers 94522-39-1P, 4,7-Methanoindene-6-d, 3a,4,5,6,7,7a-hexahydro-5-methoxy-104832-66-8P, 4,7-Methanoindan, hexahydro-5-methoxy-, exo-exo-104832-66-8P, 4,7-Methanoindan, hexahydro-5-methoxy-, exo-exo-104832-66-8P, 4,7-Methanoindan, hexahydro-5-methoxy-, exo-exo-108172-97-0P, Mercury, chloro(3a,4,5,6,7,7a-hexahydro-5-methoxy-4,7-methanoinden-6-yl)-, stereoisomers

RL: PREP (Preparation)
 (preparation of)

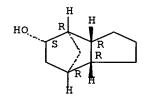
IT 10271-44-0P, 4,7-Methanoindan-5-ol, hexahydro-, exo-exo-

RL: PREP (Preparation) (preparation of)

RN 10271-44-0 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro-, (3aR,4R,5S,7R,7aR)-rel- (CA INDEX NAME)

Relative stereochemistry.



L32 ANSWER 6 OF 6 CAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1963:80843 CAPLUS Full-text

DOCUMENT NUMBER: 58:80843

ORIGINAL REFERENCE NO.: 58:13738e-h,13739a-d

TITLE: Bridged polycyclic compounds. XX. Cis stereochemistry

10/539,188

of the addition of methanol and water to

endo-trimethylenenorbornene

AUTHOR(S): Cristol, Stanley J.; Gaston, Lyle K.; Johnson, Donald

CORPORATE SOURCE: Univ. of Colorado, Boulder

SOURCE: Tetrahedron Letters (1963) 185-9

CODEN: TELEAY; ISSN: 0040-4039

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

For diagram(s), see printed CA Issue. GΙ

cf. CA 58, 10092d. H2SO4 catalyzed addition of MeOH (or of H2O in MeOH) to AΒ the isomeric endo-and exo-tri-methylenenorbornenes (I, II) gave differing ratios of endo (III) and exo (IV) ring skeleton ethers (or alcs.); e.g., I and MeOH was reported to give 14.5: 85.5 III (R = OMe) (V)-IV (R = OMe) (VI), and II with MeOH to yield 3.5: 96.5 V-VI. It is shown now that reactions with I are entirely cis-exo addns. If the endo protonated π complex (VII) reacted directly with a solvent mol. by the equivalent of trans ring opening, the addition of MeOD and D2O to I would lead to the compds. (VIII, R = OMe, OD) (IX, X). I and II lead to III and IV in such amts. that at least 80% of III from I arises via a path different from that utilized in the addition to II. If the alternative path involved VII, then at least 80% of the III isomer isolated in the deuterated series would be labeled as in VIII, and this would be true if VII were formed and gave III by isomerization to a carbonium ion (XI). If, however, VII is not involved in the reaction sequence, III might be expected to be labeled as in XII, assuming exo protonation followed by transformation to carbonium ion intermediates, coordinating with solvent from the exo side. It seemed that VIII and XII could be distinguished by differences in nuclear magnetic resonance (n.m.r.) spectra. III (R = OH) (XIII) in CHCl3 gave peaks at 6.07 (J 6.6 cycles/sec.), $7.6-9.2 \tau$. III (R = p-MeC6H4SO3, p-O2NC6H4CO2) showed similar doublets at 5.46 τ (J 6.6 cycles/sec.) and 4.90 τ (J 6.6 cyles/sec.). It appeared that the principal coupling was due to interaction between H and the C-3 H eclipsed with it, and that this coupling would be observed in XII, but would not appear in VIII. Treatment of 5,6-endo-trimethylene-2,3-exo-epoxynorbornane with LiAlD4 gave VIII (R = OH), showing a broad singlet at 6.02 τ ; p-O2NC6H4CO2 derivative m. 123-6°, n.m.r. singlet at 4.89 τ . Addition of B2D6 to I followed by oxidation gave XII (R = OH), n.m.r. doublet at 6.09 τ (J 6.9 cycles/sec.); p-O2NC6H4CO2 derivative m. 128-9°, doublet at 4.99 τ (J 7.0 cycles/sec.). I (8.04 g. containing 9.7% II and 1.5% dicyclopentadiene), 5.5 ml. D2SO4, 2.2 ml. D2O, and 46 ml. MeOD refluxed 10 hrs. with stirring and treated with 10 g. Na2SO4, poured into H2O, and extracted with Et2O, the Et2O evaporated, and the residue [5.2% I, 10.3% V, 68.7% VI, 2.6% XIII, and 13.2% IV (R = OH) (XIV) by gas phase chromatography] chromatographed on Al203, eluted with pentane, and the ether mixture separated on a 2 m. (3/8 in.) 30% Carbowax 1540-Chromosorb column at 123° in 88 ml.He/min. gave I, XIV, and V in 26, 106, and 136 min. V showed a fairly sharp doublet at 6.74. τ (J.6.7) and the spectrum indicated that less than 10% of IX could be present as a contaminant in the XIII (R =OMe) produced. Further elution with Et2O gave the alc. fractions which were converted to p-nitrobenzoates inseparable into isomers. The esters reduced with LiAlH4, chromatographed on Al2O3, and separated by gas chromatography on a 1 m. (1/4 in.) 30% Carbowax 1540 column at 130° with 88 ml. He/min. gave XIV and XIII in 66 and 86 min., resp. XIII showed a doublet at 6.08 τ (J 6.6 cycles/sec.). Less than 10% of X could be present in the XII (R = OH) produced. Clearly cis-exo addition of both the H and R portions of the addenda occurs and the endo-protonated complex VII is excluded as a significant intermediate in the addition reaction to I leading to III. It is suggested that at least one of the intermediates involved is a classical ion

such as XI, or that a cis addition 4-center transition state is involved, in which no carbonium ion is developed.

CC 32 (Physical Organic Chemistry)

IT 2198-08-5 2415-40-9 2628-89-9 **13380-89-7** 19398-83-5 68628-62-6 **96655-49-1** 97811-11-5

(Derived from data in the 7th Collective Formula Index (1962-1966))

IT 13380-89-7 96655-49-1

(Derived from data in the 7th Collective Formula Index (1962-1966))

RN 13380-89-7 CAPLUS

CN 4,7-Methano-1H-inden-5-ol, octahydro- (CA INDEX NAME)

RN 96655-49-1 CAPLUS

CN 4,7-Methanoindan-6-d-5-ol, hexahydro- (7CI) (CA INDEX NAME)

=> d his nofil

L5

(FILE 'HOME' ENTERED AT 15:58:41 ON 15 JAN 2008)

FILE 'CAPLUS' ENTERED AT 15:58:51 ON 15 JAN 2008 E US2005-539188/APPS

L1 1 SEA ABB=ON PLU=ON US2005-539188/AP

SEL RN

D SCA

E DEUTERATION/CT

E E3+ALL

L2 2077 SEA ABB=ON PLU=ON DEUTERATION+PFT/CT

E DEUTERATION CATALYSTS+ALL/CT

L3 287 SEA ABB=ON PLU=ON DEUTERATION CATALYSTS+PFT,NT/CT

L4 2148 SEA ABB=ON PLU=ON L2 OR L3

FILE 'REGISTRY' ENTERED AT 16:00:21 ON 15 JAN 2008

FILE 'CAPLUS' ENTERED AT 16:00:34 ON 15 JAN 2008

TRA PLU=ON L4 1- RN: 22168 TERMS

FILE 'REGISTRY' ENTERED AT 16:01:30 ON 15 JAN 2008

L6 22168 SEA ABB=ON PLU=ON L5

L7 952 SEA ABB=ON PLU=ON L6 AND (PD OR PT OR RH OR RU OR NI OR CO)/ELS

FILE 'CAPLUS' ENTERED AT 16:03:17 ON 15 JAN 2008

10/539,188 January 15, 2008

```
L8
        154222 SEA ABB=ON PLU=ON L7(L)CAT+NT/RL
L9
            247 SEA ABB=ON PLU=ON L8 AND L4
            115 SEA ABB=ON PLU=ON L2 AND L3 AND L8
L10
     FILE 'REGISTRY' ENTERED AT 16:04:58 ON 15 JAN 2008
L11
                STR
     FILE 'CAPLUS' ENTERED AT 16:10:55 ON 15 JAN 2008
                S L11
     FILE 'REGISTRY' ENTERED AT 16:11:21 ON 15 JAN 2008
L12
             50 SEA SUB=L6 SSS SAM L11
     FILE 'CAPLUS' ENTERED AT 16:11:21 ON 15 JAN 2008
L13
             68 SEA ABB=ON PLU=ON L12
     FILE 'REGISTRY' ENTERED AT 16:11:26 ON 15 JAN 2008
L14
             50 SEA SUB=L6 SSS SAM L11
L15
           2492 SEA SUB=L6 SSS FUL L11
L16
           1209 SEA ABB=ON PLU=ON L15 AND D/ELS
     FILE 'CAPLUS' ENTERED AT 16:12:27 ON 15 JAN 2008
L17
           897 SEA ABB=ON PLU=ON L16(L) PREP+NT/RL
L18
            40 SEA ABB=ON PLU=ON L17 AND L9
L19
             22 SEA ABB=ON PLU=ON L17 AND L10
             1 SEA ABB=ON PLU=ON L1 AND L1
L20
     FILE 'CAPLUS' ENTERED AT 16:13:16 ON 15 JAN 2008
                D OUE L19
                D L19 IBIB ABS HITIND HITSTR TOT
                SEL RN L1
     FILE 'REGISTRY' ENTERED AT 16:15:20 ON 15 JAN 2008
L21
             49 SEA ABB=ON PLU=ON (7440-16-6/BI OR 108-93-0/BI OR 13380-89-7/
                BI OR 63870-91-7/BI OR 725242-29-5/BI OR 7440-05-3/BI OR
                7440-06-4/BI OR 7440-18-8/BI OR 79-31-2/BI OR 106-35-4/BI OR
               108-94-1/BI OR 110-43-0/BI OR 123-19-3/BI OR 127-09-3/BI OR
               13380-94-4/BI OR 14044-94-1/BI OR 18153-61-2/BI OR 21273-02-9/B
               I OR 3385-61-3/BI OR 350820-09-6/BI OR 497-38-1/BI OR 51209-49-
               5/BI OR 53481-06-4/BI OR 543-49-7/BI OR 5536-61-8/BI OR
               55935-44-9/BI OR 589-55-9/BI OR 64118-21-4/BI OR 666-52-4/BI
               OR 67-64-1/BI OR 725242-18-2/BI OR 725242-19-3/BI OR 725242-21-
               7/BI OR 725242-22-8/BI OR 725242-23-9/BI OR 725242-24-0/BI OR
               725242-25-1/BI OR 725242-26-2/BI OR 725242-27-3/BI OR 725242-28
               -4/BI OR 725242-30-8/BI OR 725242-31-9/BI OR 725242-32-0/BI OR
               7440-02-0/BI OR 7440-48-4/BI OR 7789-20-0/BI OR 78-93-3/BI OR
               79-41-4/BI OR 91468-78-9/BI)
L22
            10 SEA ABB=ON PLU=ON L21 AND NR>2
               D SCA
             2 SEA ABB=ON PLU=ON L22 AND ?DECAN?/CNS
L23
               D SCA
             1 SEA ABB=ON PLU=ON L23 AND C10H16O/MF
L24
               D
               SEL RN
L25
             3 SEA ABB=ON PLU=ON 13380-89-7/CRN
               D SCA
    FILE 'CAPLUS' ENTERED AT 16:17:52 ON 15 JAN 2008
L26
             2 SEA ABB=ON PLU=ON L25
               D OUE L26
```

D L26 IBIB ABS HITSTR TOT

L27 L28	FILE 'REGISTRY' ENTERED AT 16:18:52 ON 15 JAN 2008 STR 13380-89-7 52 SEA FAM FUL L27
	FILE 'CAPLUS' ENTERED AT 16:19:11 ON 15 JAN 2008
L29	133 SEA ABB=ON PLU=ON L28
L30	168165 SEA ABB=ON PLU=ON ?DEUTER?
L31	4 SEA ABB=ON PLU=ON L29 AND L30
L32	6 SEA ABB=ON PLU=ON L31 OR L26
	FILE 'CAPLUS' ENTERED AT 16:19:51 ON 15 JAN 2008
	D QUE L32
	D L32 IBIB ABS HITIND HITSTR TOT